

§3 立体化学

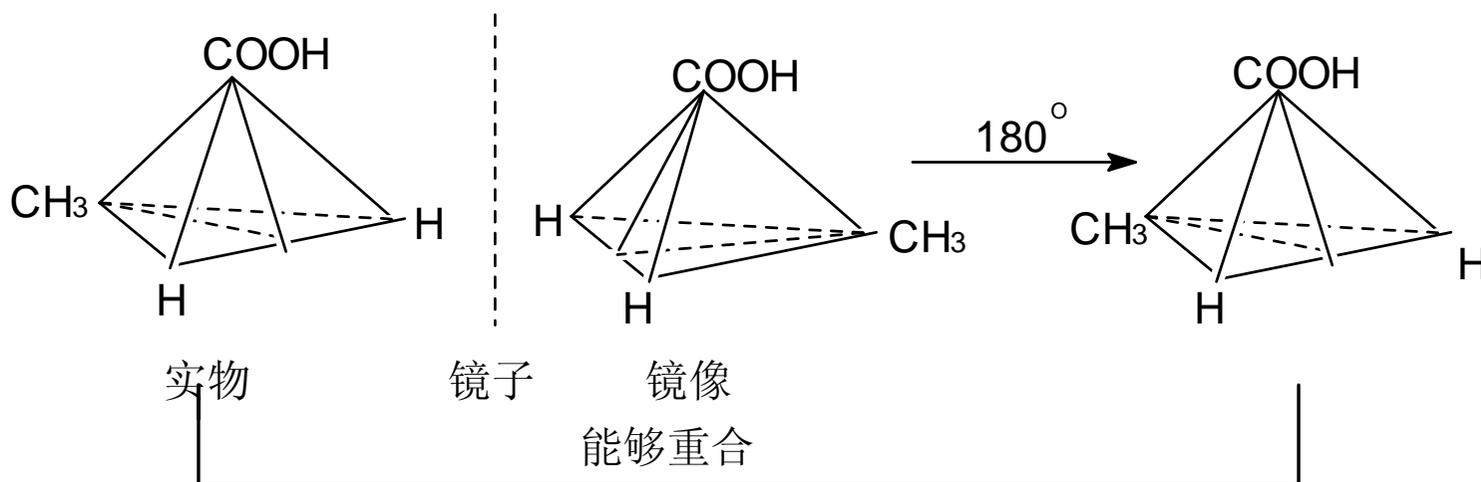
立体化学基础

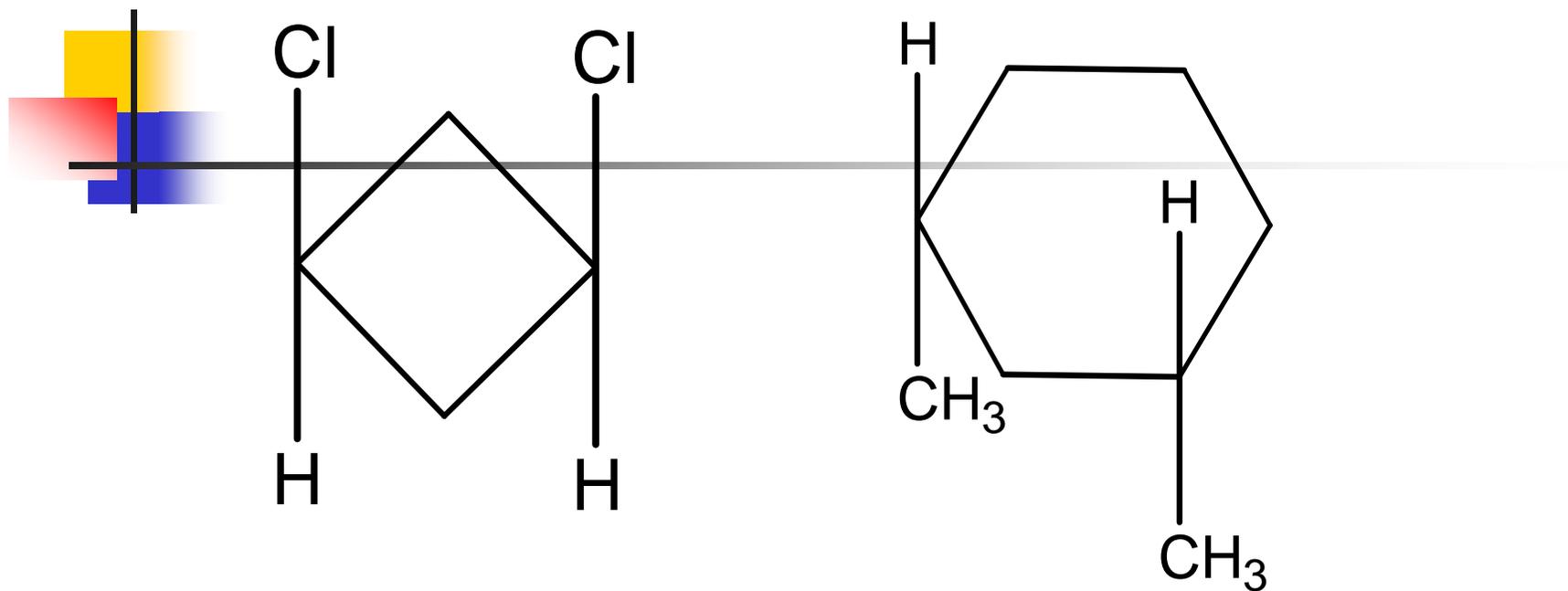
§ 3.1 对称因素与手性分子

1. 对称因素

A. 对称面

分子中有一平面将分子一分两半，两部分互为镜像关系，该平面称为对称面，用 σ 表示。

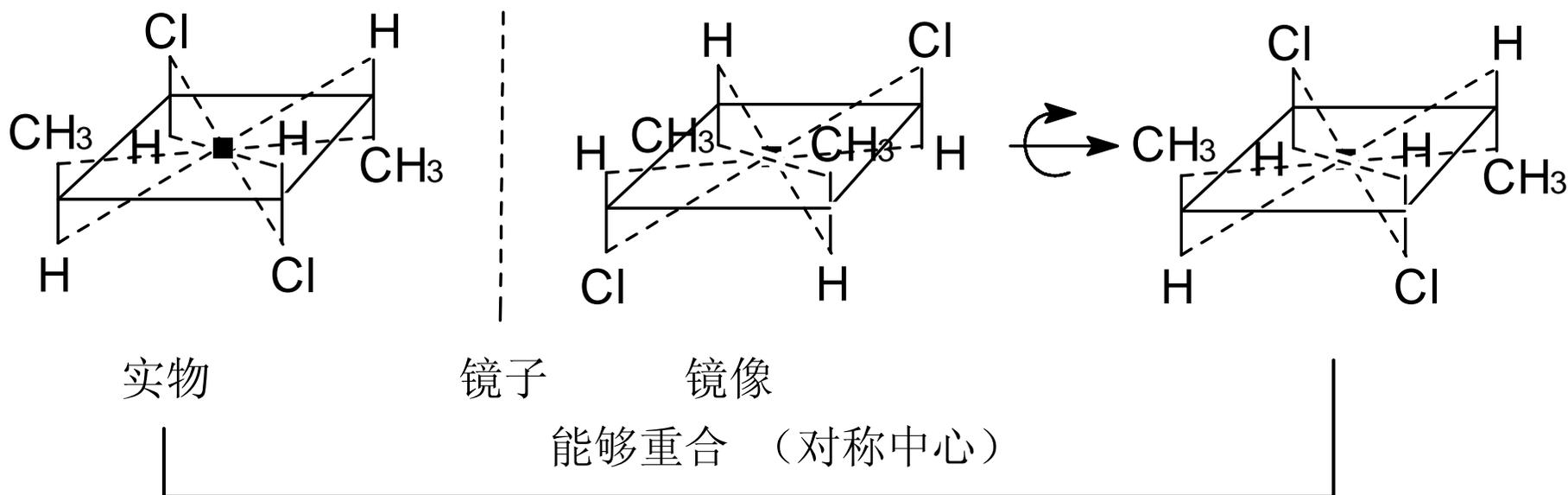


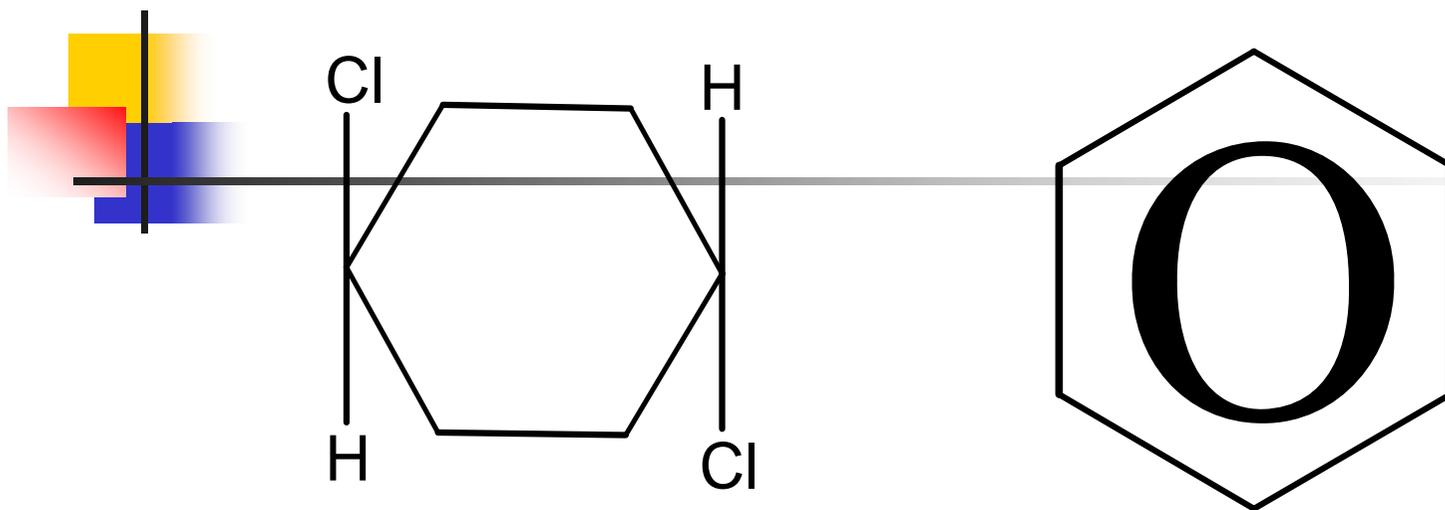


结论：如果分子中存在对称面，那么该分子的镜像可以与实物重合。

B. 对称中心

在分子中可找到某一个点，通过该点任意向两端作射线，距离该点相等的距离有相同的原子或基团，则该点称为该分子的对称中心。用 I 表示。





结论：如果分子中存在对称中心，该分子的镜像可以与实物重合。

C.对称轴

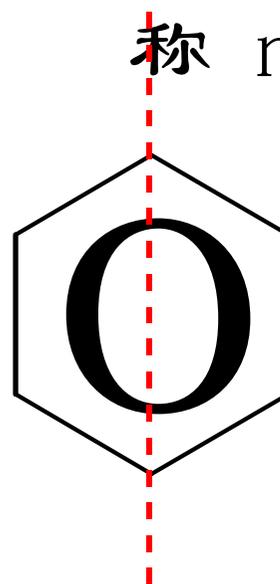
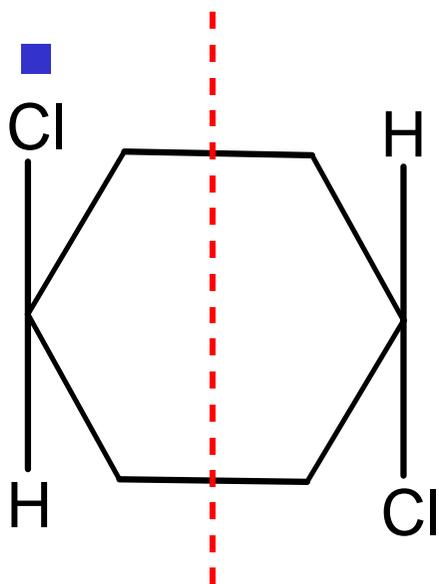
在分子中可找到某条直线，分子绕该直线旋转一定角度后可与原分子重合，则该直线称为该分子的对称轴。用 C_n 表示。其中：

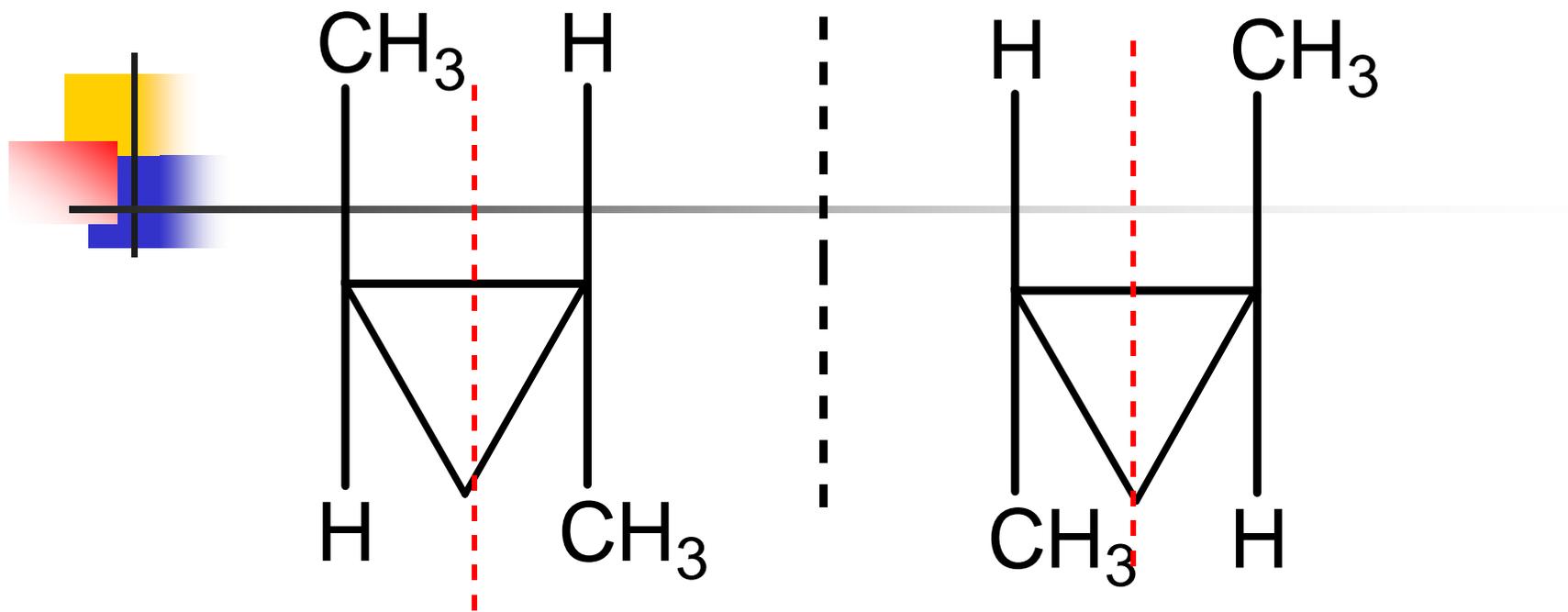
$$360^\circ$$

$$n = \frac{\quad}{\quad}$$

旋转的角度

称 n 重对称轴





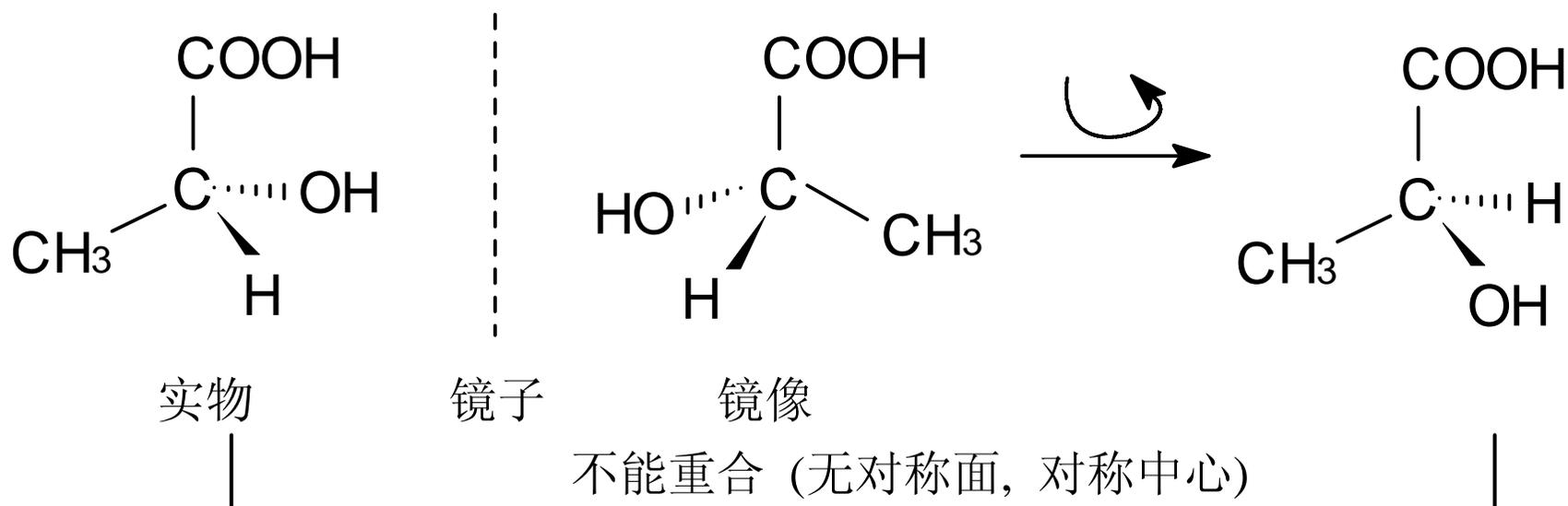
结论：虽然分子中存在C₂对称轴，但该分子的镜像却不可以与实物重合。

2. 手性及手性分子

■ 手性: 不对称性

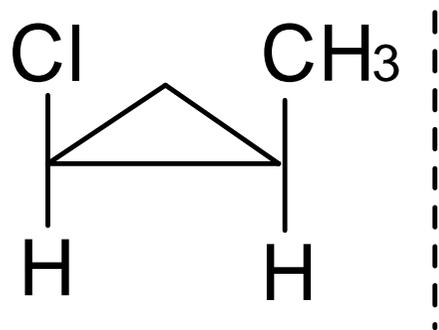
■ A. 手性分子

■ a. 实例: 乳酸

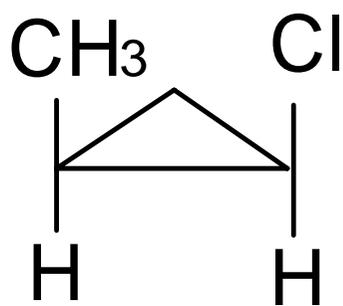


1-氯-2-甲基环丙烷

cis

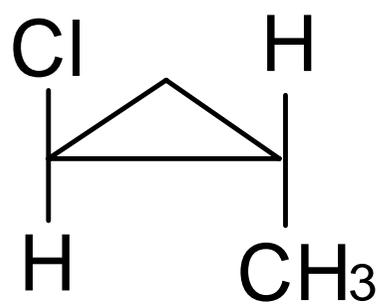


实物

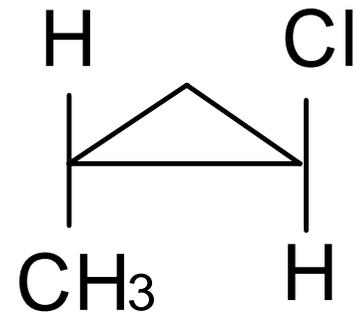


镜像

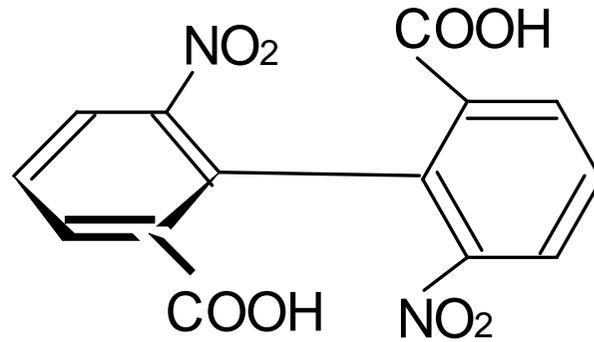
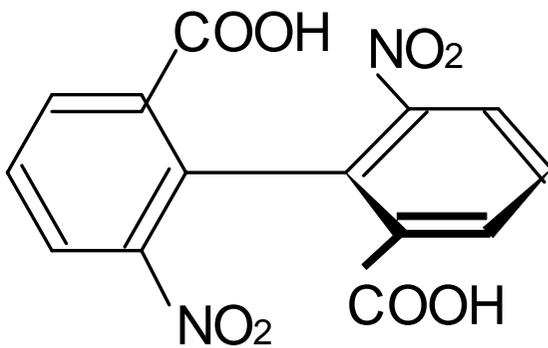
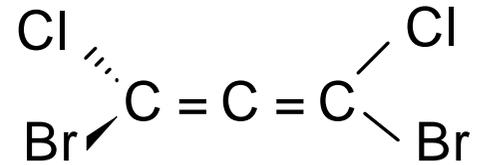
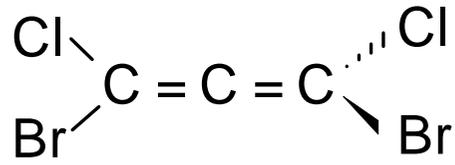
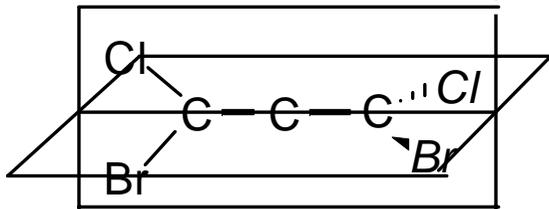
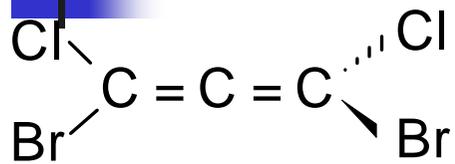
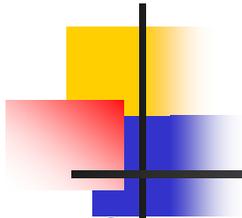
trans

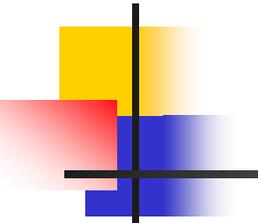


实物



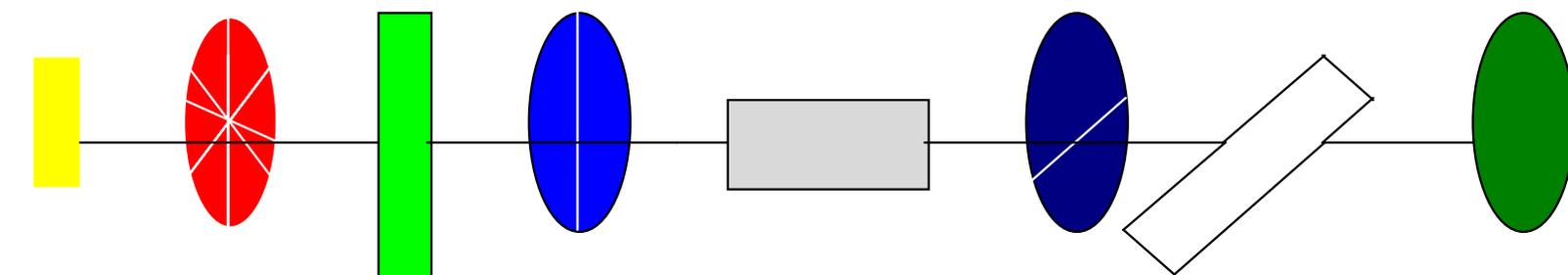
镜像



- 
- b. **手性分子**：实物与镜像不能重合的分子是手性分子，即-不对称的分子。
 - 实物与镜像的关系为对映关系。
实物与镜像互为**对映(异构)体**。

- c. **手性分子产生的条件**：既无对称面
也无对称中心

8. 手性分子的物理性质——旋光性



光源

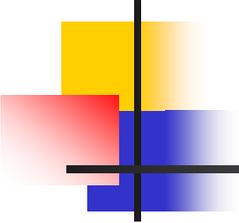
尼可尔
棱镜
(起偏器)

手性分子
溶液

尼可尔
棱镜
(检偏器)

读数
右旋 (+)
左旋 (-)

旋光测定示意图



■ 比旋光度 $[\alpha]_{\lambda}^t = \alpha / (c l)$

α 测得的旋光度 + 或 -

c 溶液浓度 (g/ml)

l 旋光管长度 (10cm)

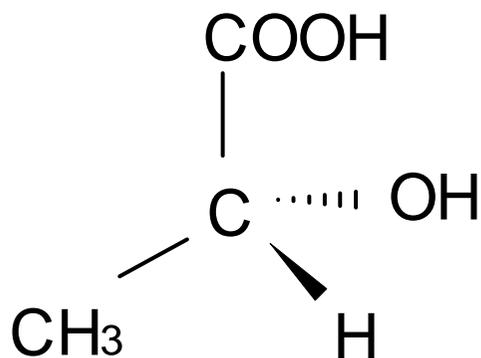
t 温度 通常为25 °C

■ λ 波长 通常为Na光源 (5893Å⁰)

■ 即D光源

§ 3.2 构型及构型的平面表达

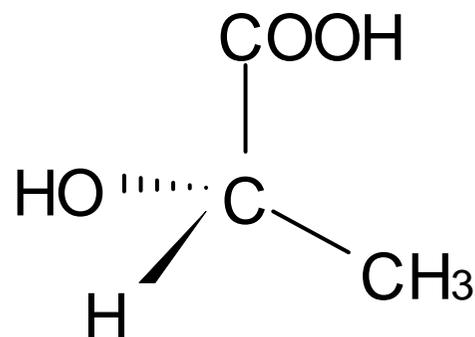
- 1. 构型：分子中的原子或基团在空间的特定排布方式称为构型



构型 I

$$[\alpha]_{\lambda}^t = +3.8^{\circ}$$

实物



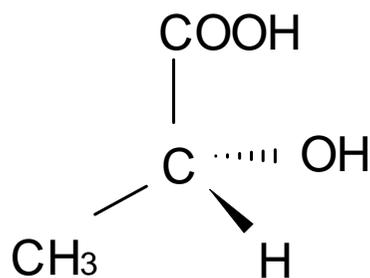
构型 II

$$[\alpha]_{\lambda}^t = -3.8^{\circ}$$

镜像

2. 对映(异构)体及外消旋体

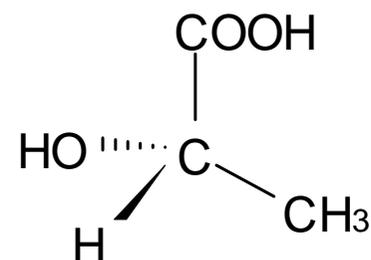
■ A. 对映体:



构型 I

$$[\alpha]_{\lambda}^t = +3.8^{\circ}$$

实物



构型 II

$$[\alpha]_{\lambda}^t = -3.8^{\circ}$$

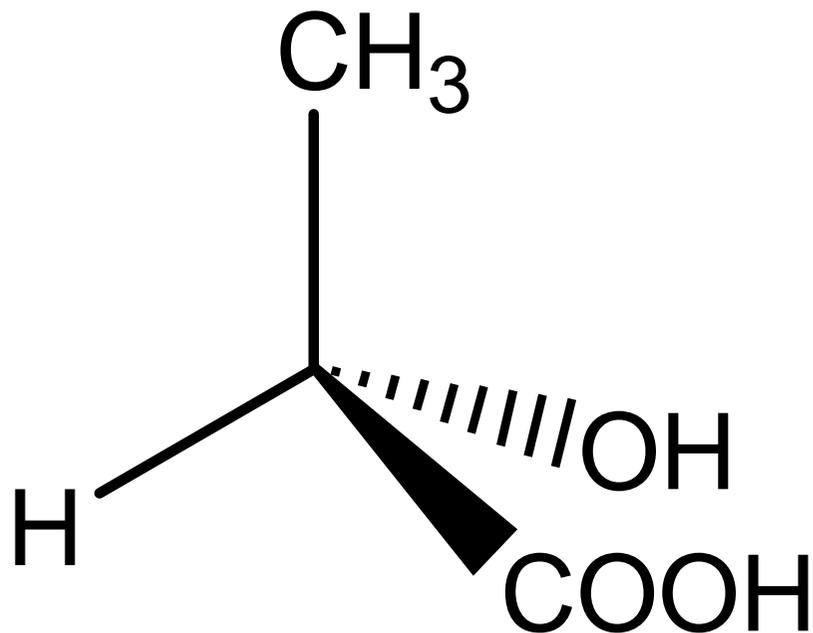
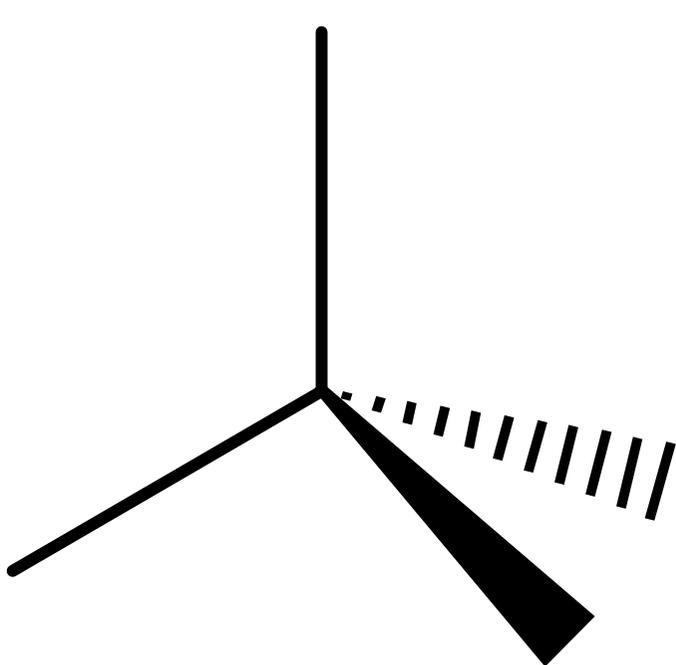
镜像

■ B. 外消旋体:

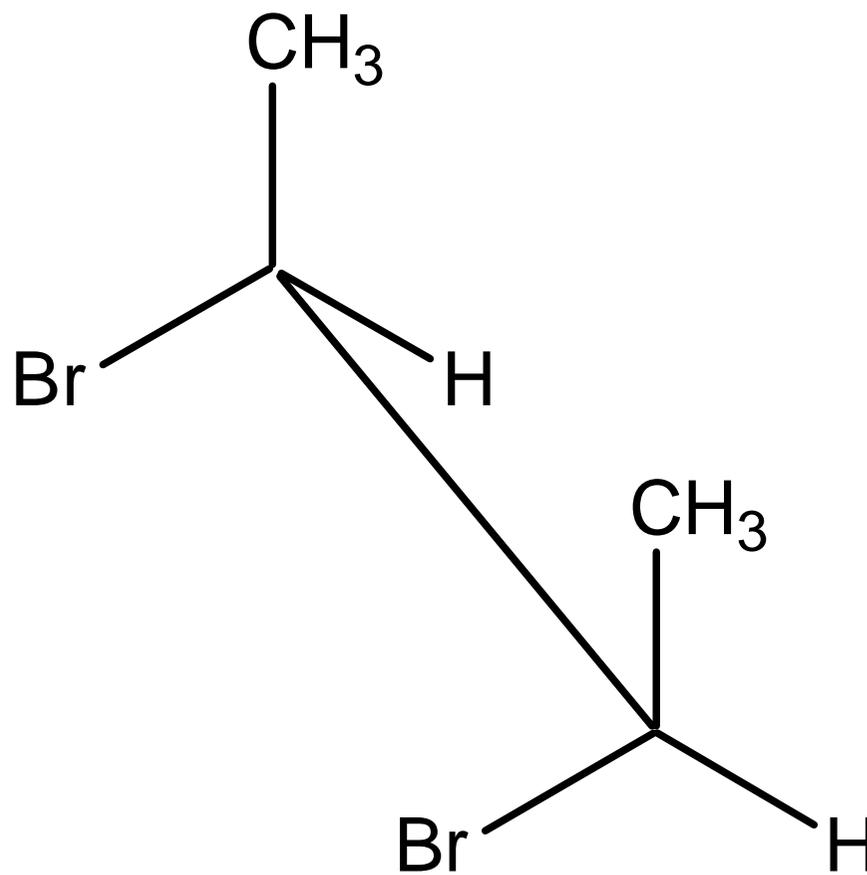
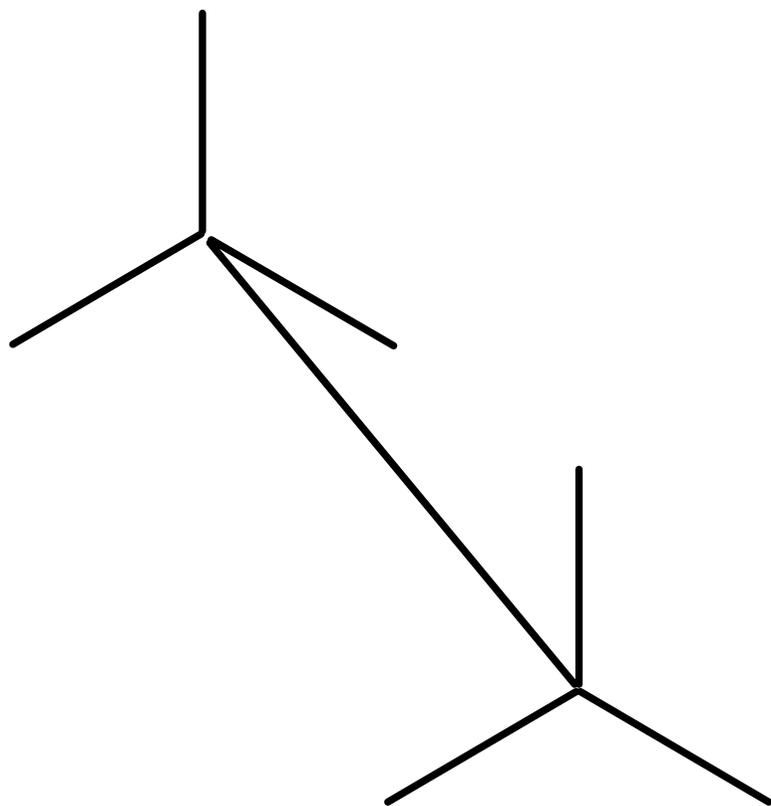
■ 等量对映体的混合物

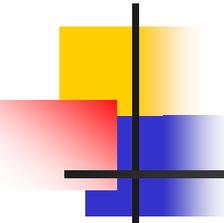
3. 构型的平面表达

- A, 伞形式:

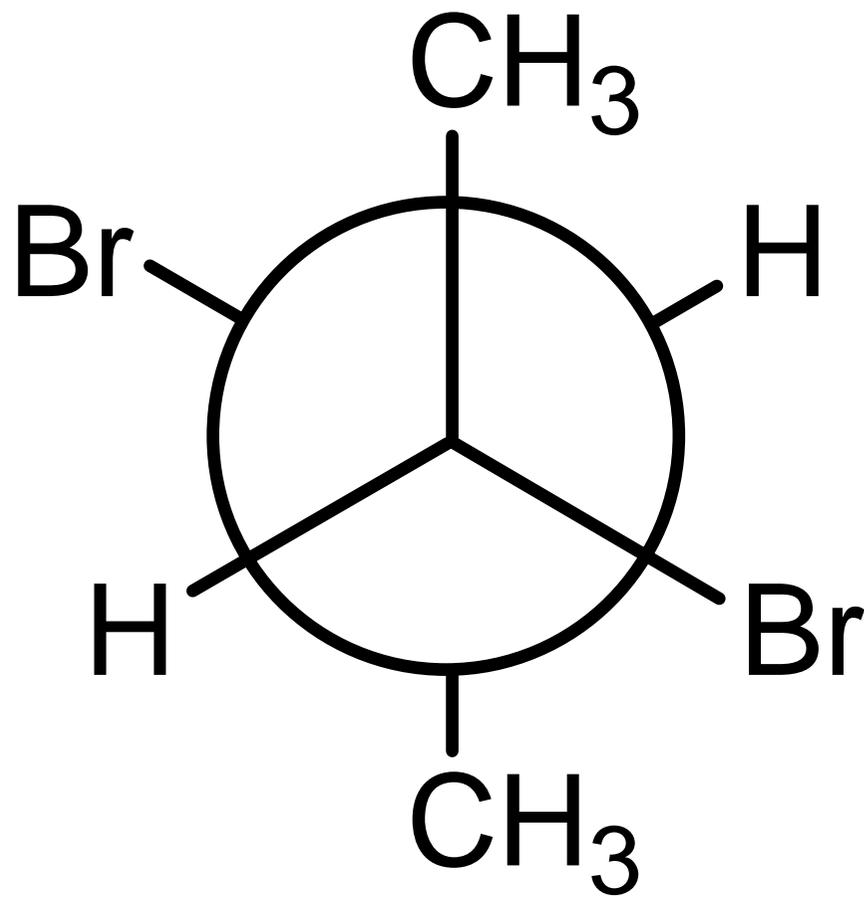
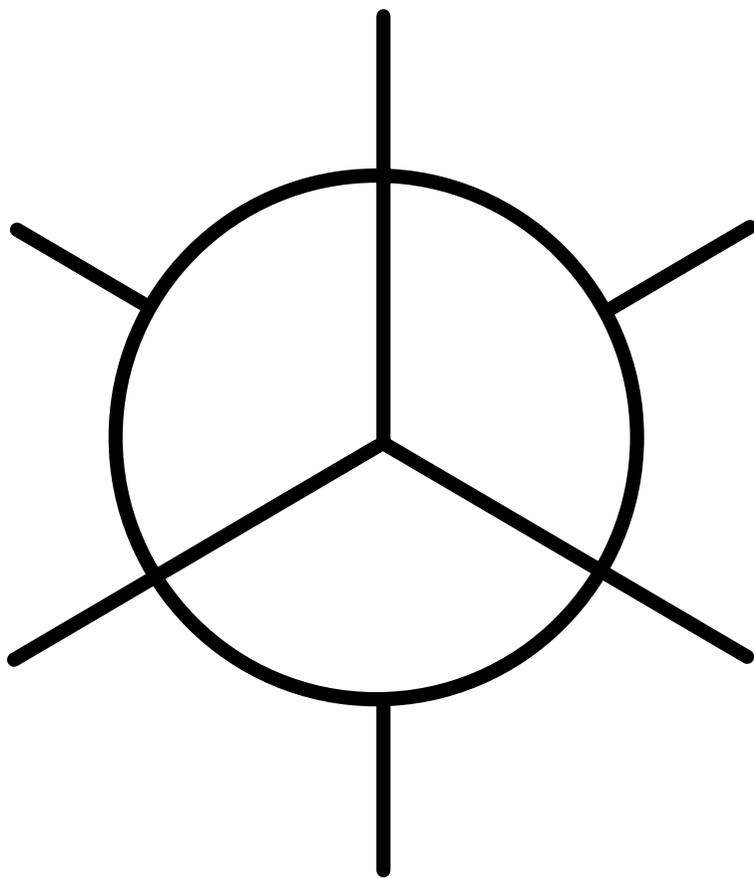


8. 锯架式:



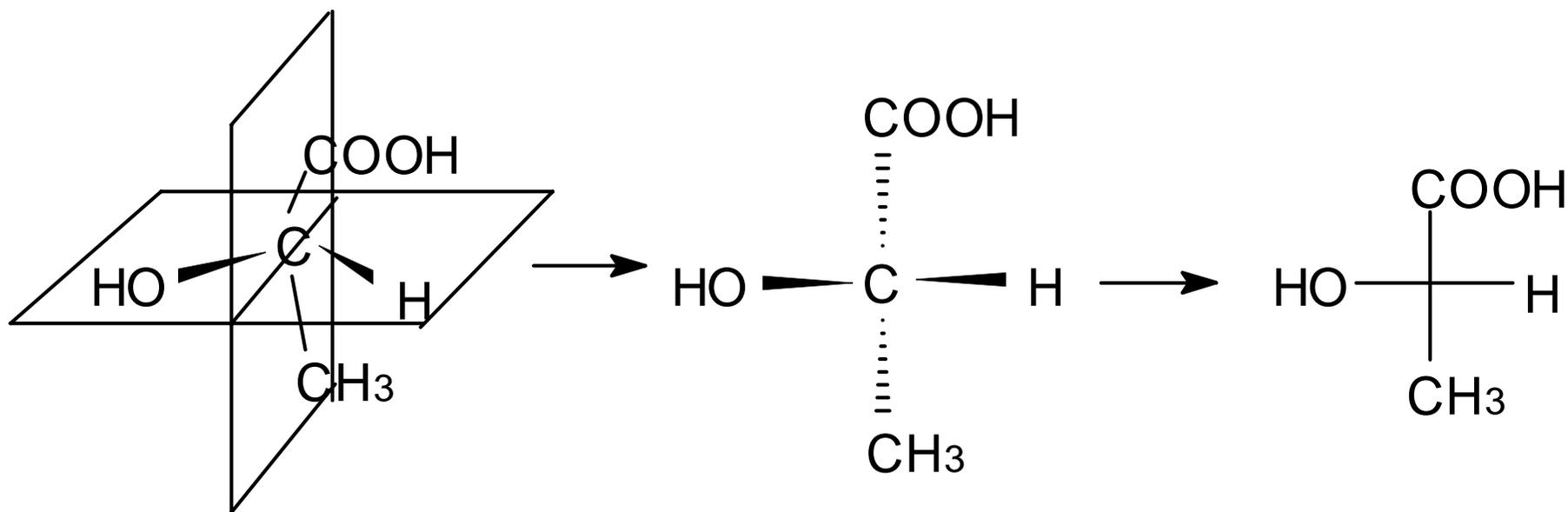


C. 纽曼式:



D, Fischer投影式

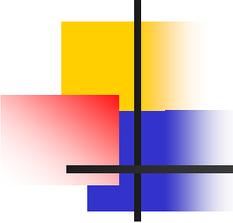
a, Fischer投影式:



1) 十字型交叉点为手性碳;

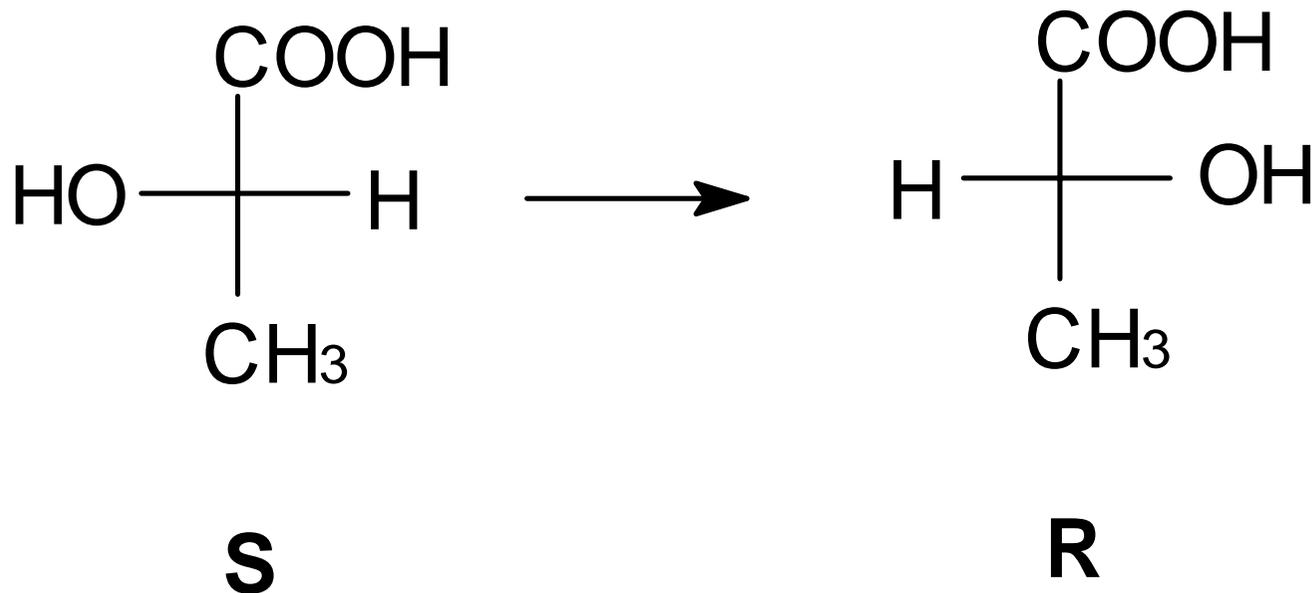
2) 横线表示向前; 竖线表示向后

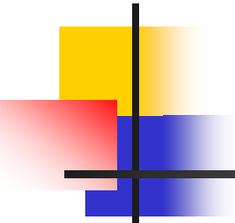
3) 一般把含碳基团写在竖线上, 氧化态高的写在上

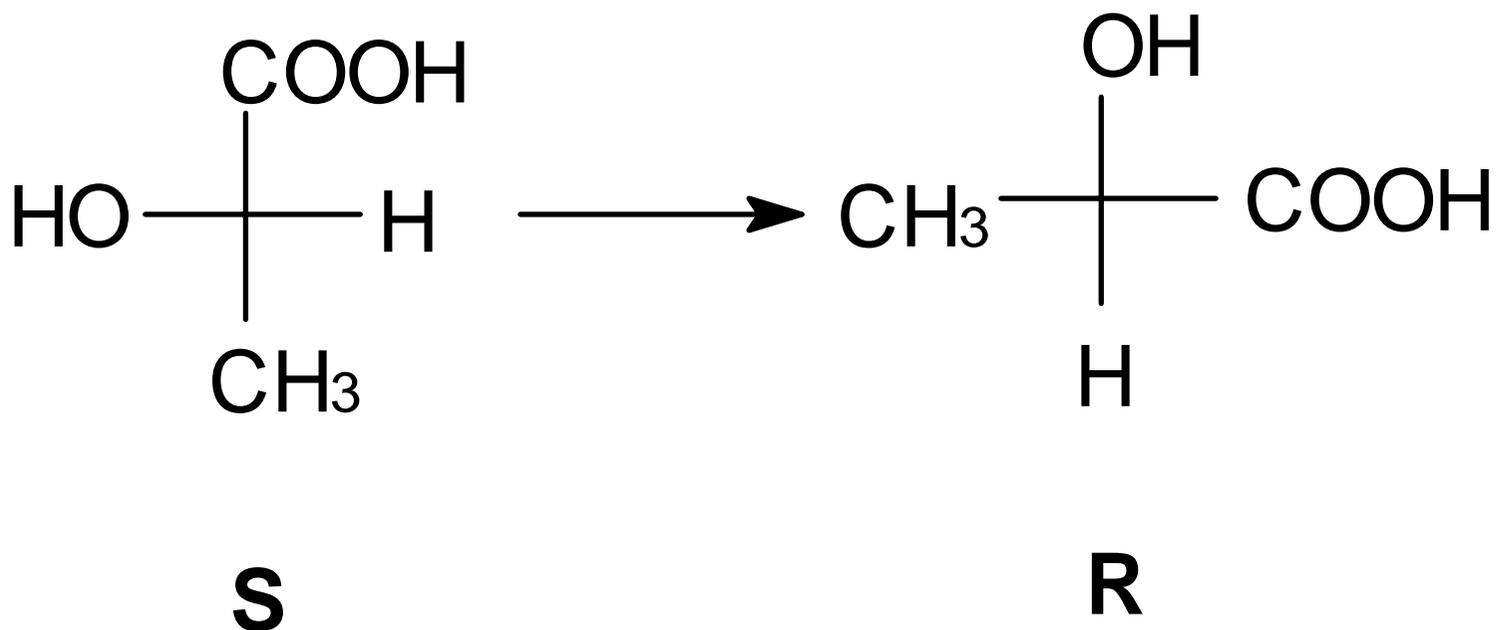


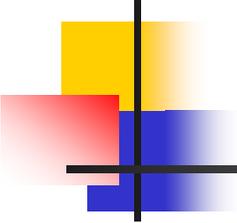
6. Fischer投影式的规则

- 1) 交换两个基团，变为对映体

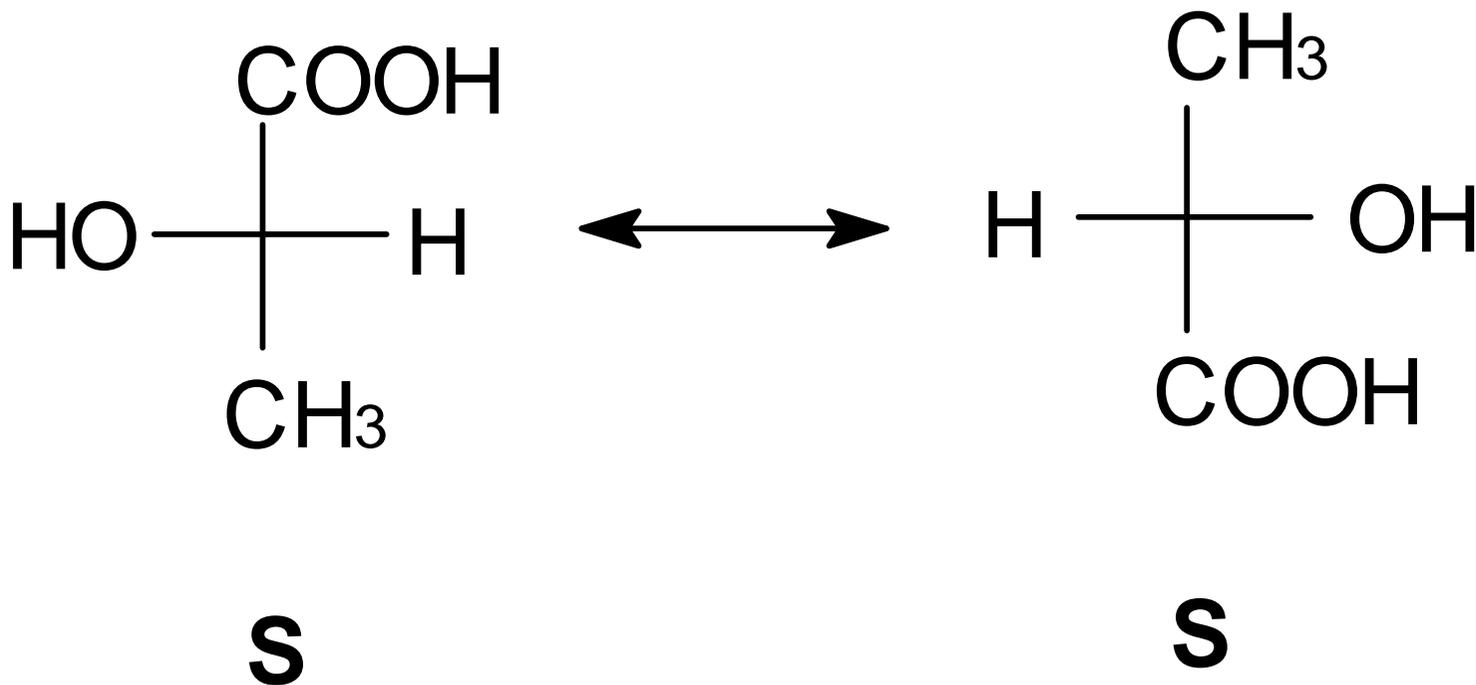


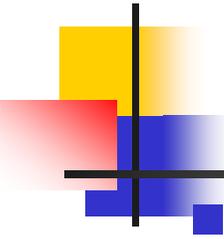
- 
- 2) 面内旋转 90° ，或面外翻转 180° 变为对映体



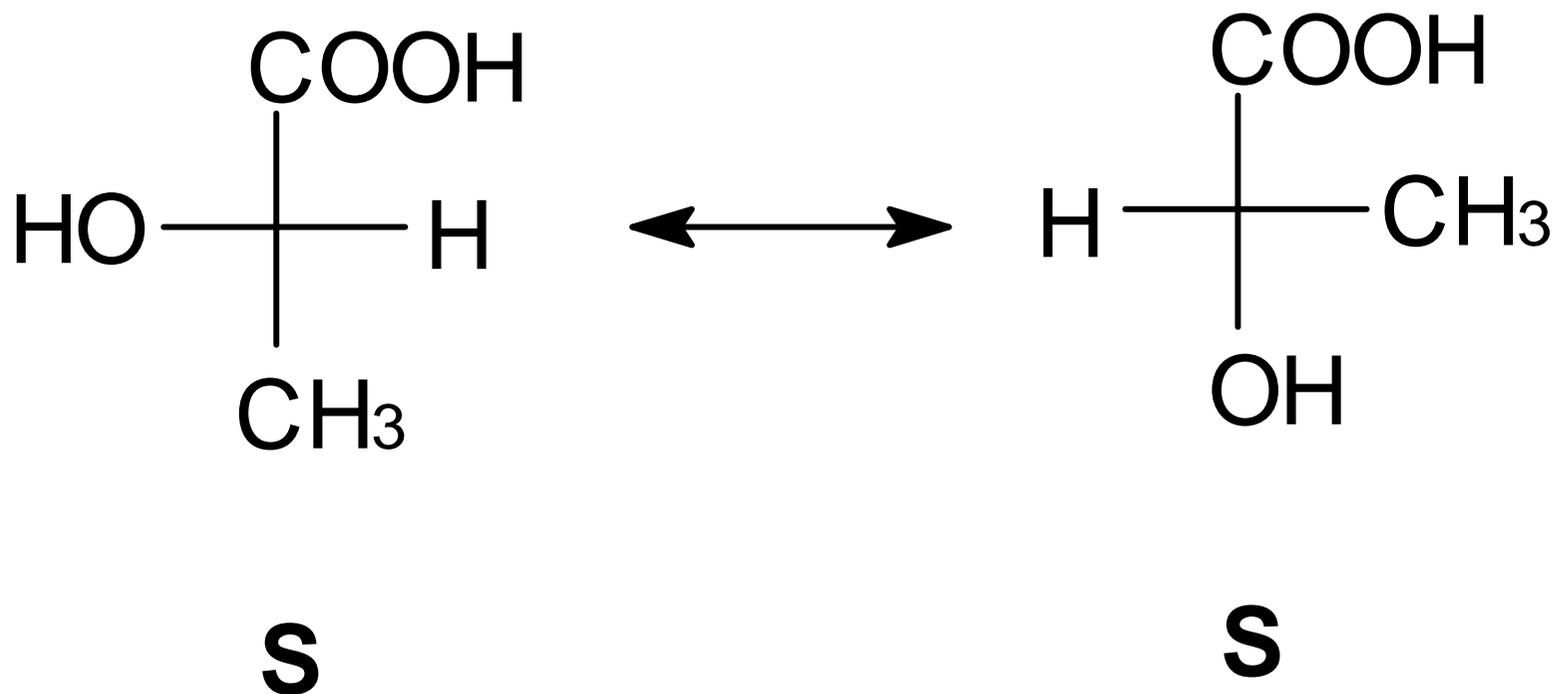


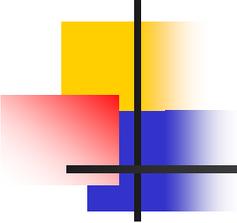
■ 3) 面内旋转 180° , 构型不变





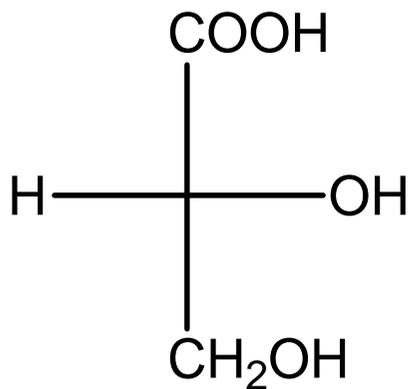
4) 固定一个基团不动，按次序交换另三个基团，构型不变；



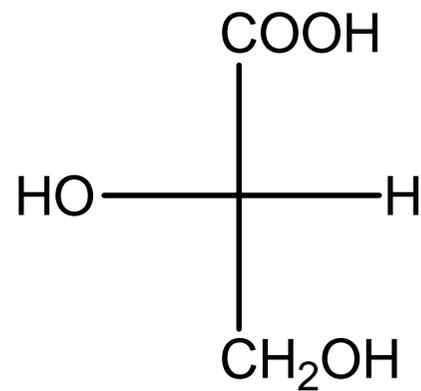


§ 3.3 构型的标识

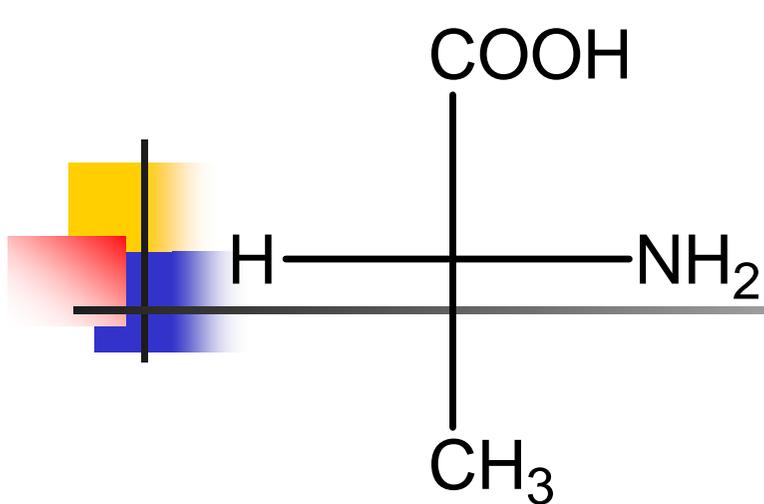
1. D/L 构型标识法：



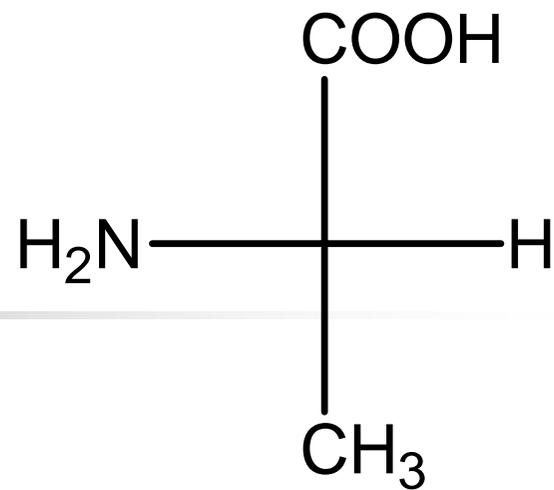
D-甘油酸



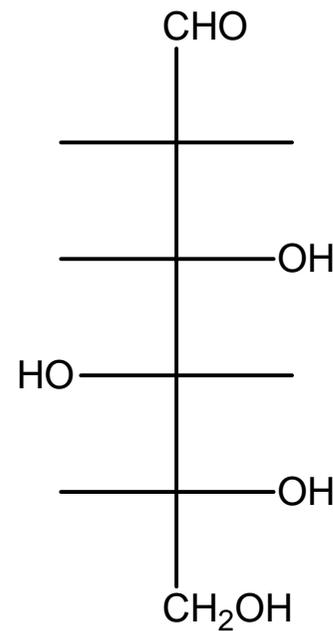
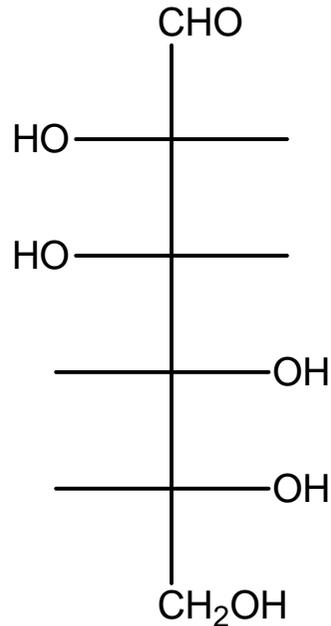
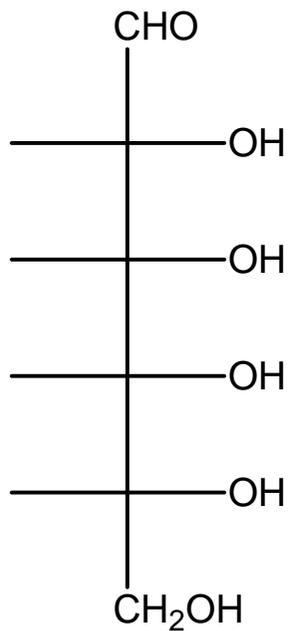
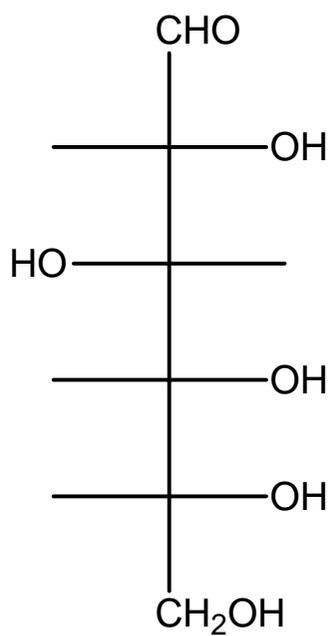
L-甘油酸

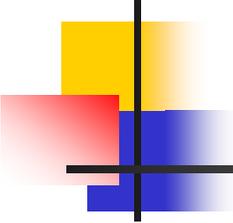


D-丙氨酸

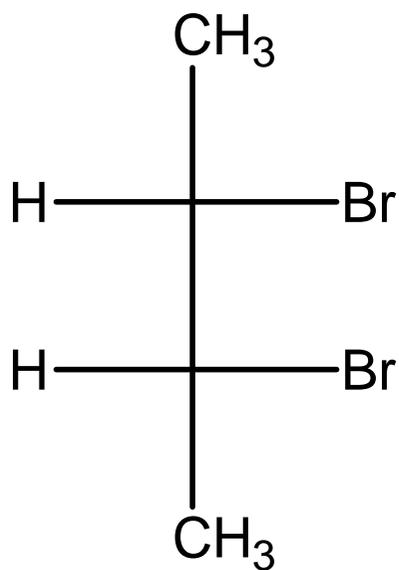


L-丙氨酸

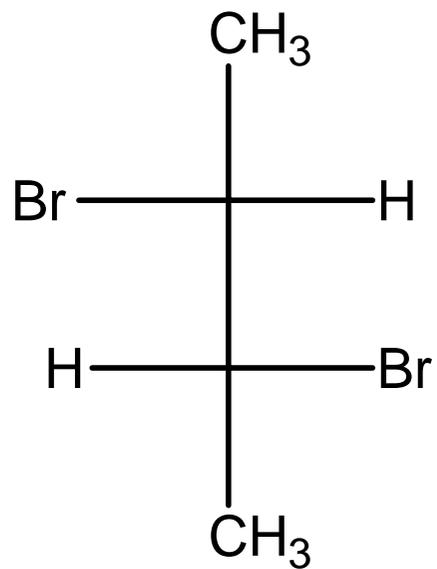




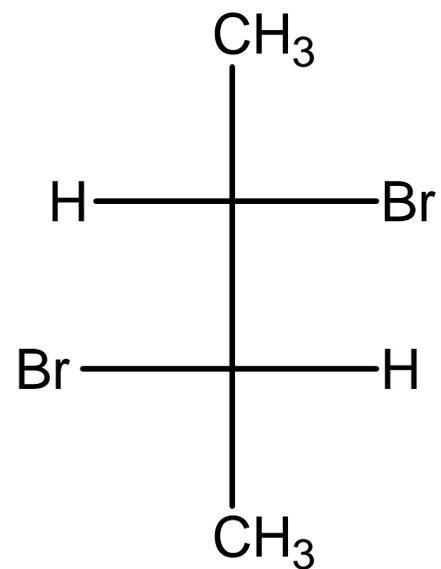
2. 赤式/苏式 构型标识法：

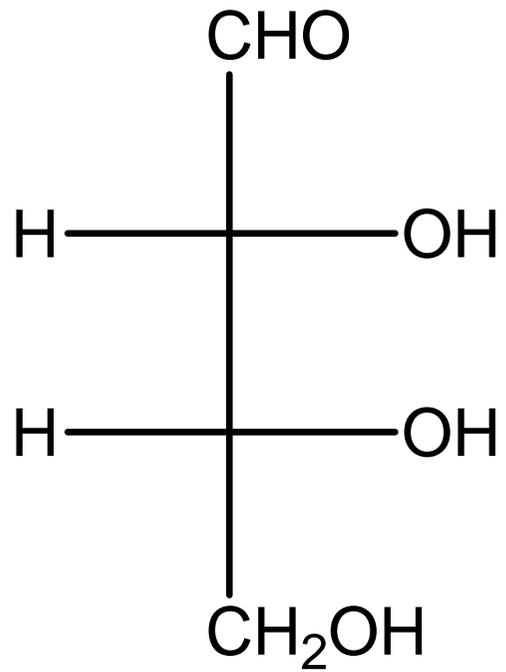
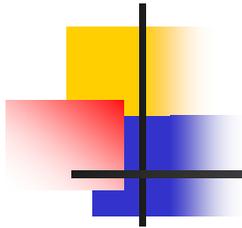


赤式

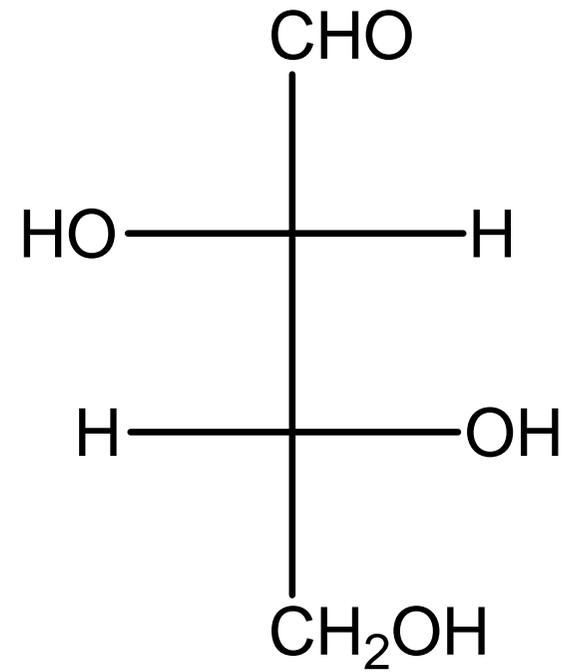


苏式

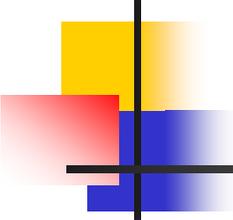




赤藓糖



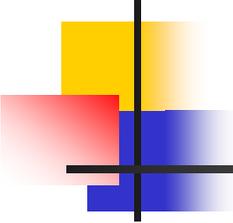
苏阿糖



3. R/S 构型标识法 (IUPAC) :

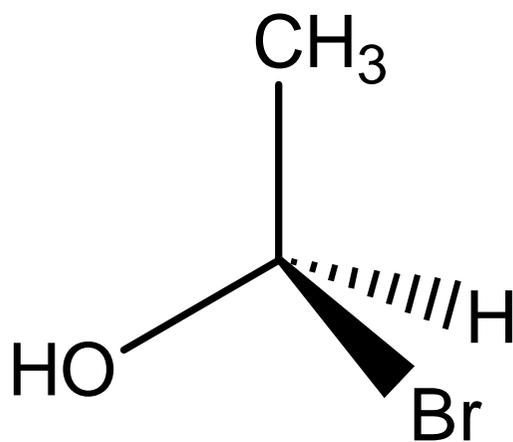
A.规定：(1)把基团按一定的定序规则定序。

(2)朝向最小基团，按由大到小看另三个基团，若为顺时针则定义该手性碳的构型为 R 构型；反之，若为逆时针则定义该手性碳的构型为 S 构型。

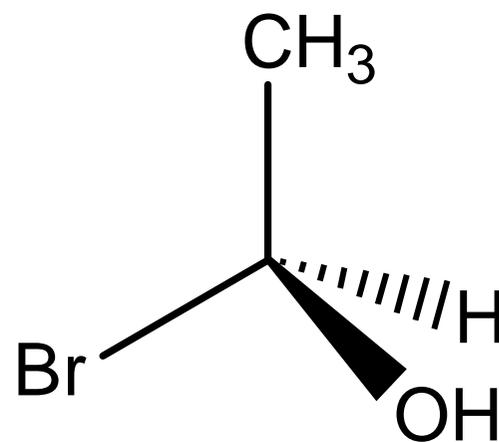


如：

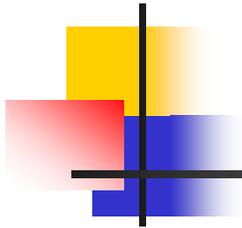
若按一定的定序规则： $\text{Br} > \text{OH} > \text{CH}_3 > \text{H}$



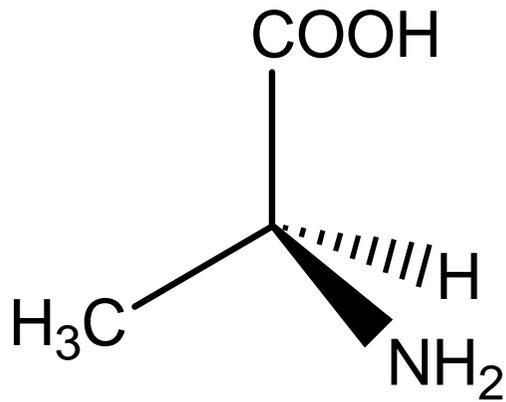
R



S

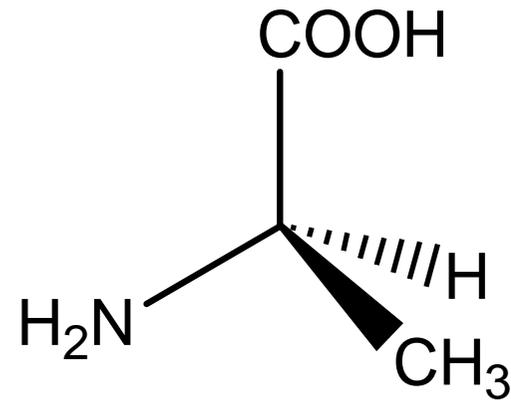


若按一定的定序规则： $\text{NH}_2 > \text{COOH} > \text{CH}_3 > \text{H}$



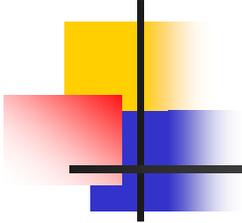
S-丙氨酸

L-丙氨酸

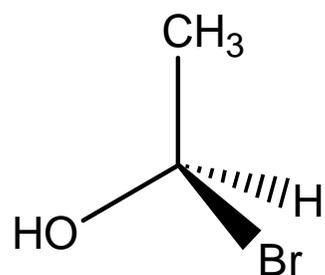


R-丙氨酸

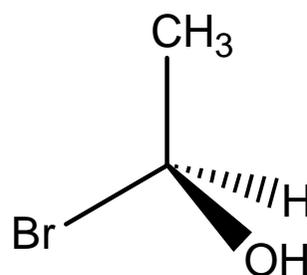
D-丙氨酸



若按一定的定序规则: $\text{Br} > \text{OH} > \text{CH}_3 > \text{H}$

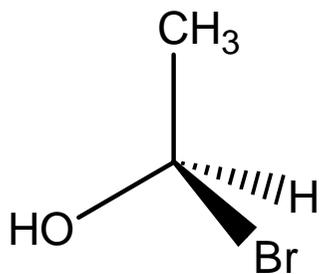


R

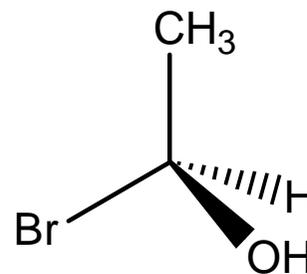


S

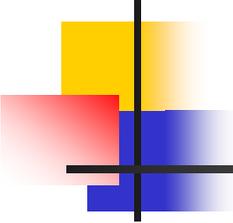
若按一定的定序规则: $\text{OH} > \text{Br} > \text{CH}_3 > \text{H}$



S



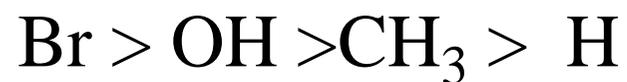
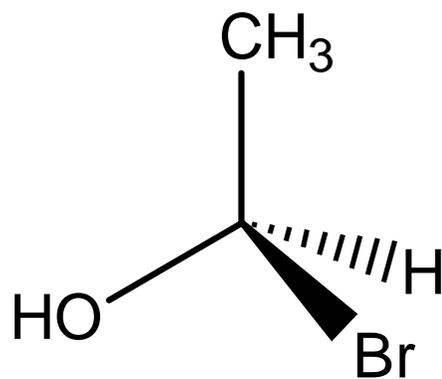
R

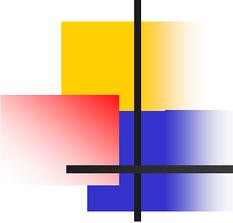


B. 定序规则：

1, 若一个手性碳上连有四个不同的原子, 则原子序数大的优先。

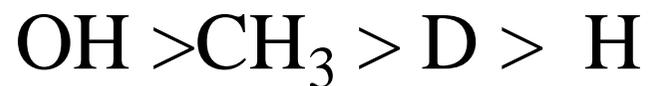
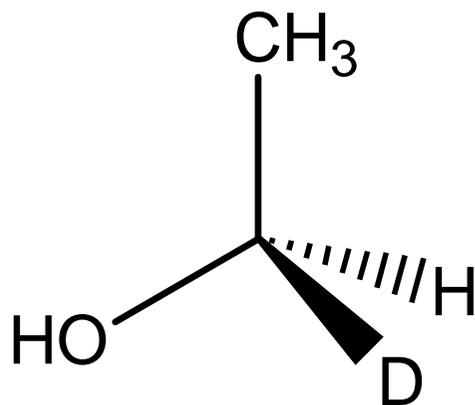
如：



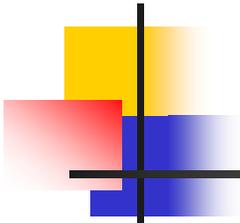


2, 若一个手性碳上连有原子序数相同的原子,
则原子量大的优先。

如:

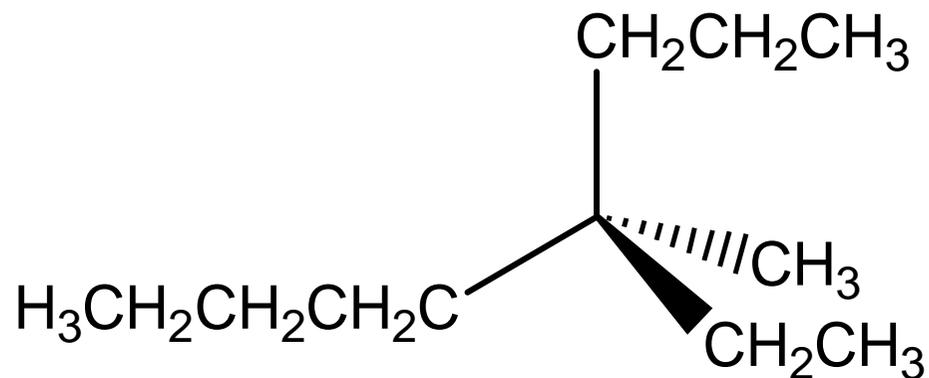


R



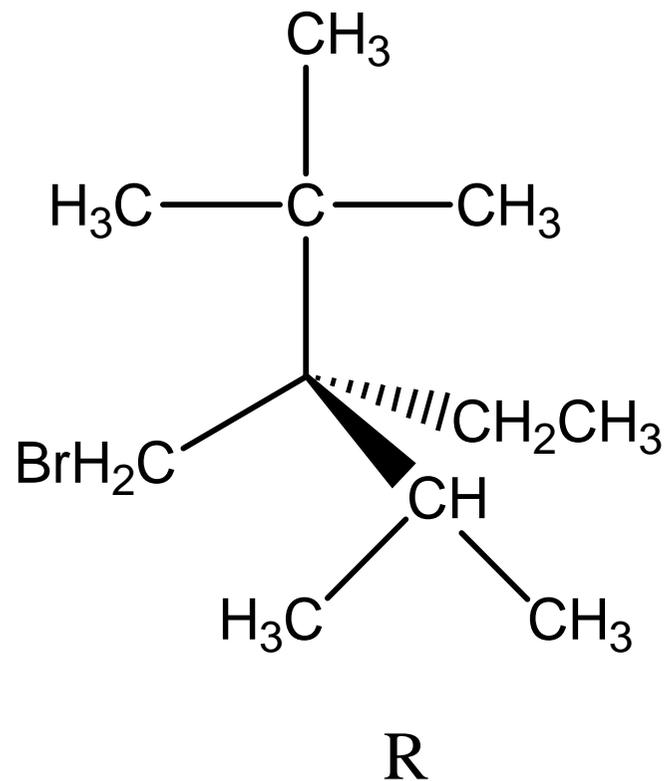
3, 若一个手性碳上连有原子序数相同的原子,
且原子量也相同, 则按规则 1, 2, 次序地比
较下一个所连的原子。

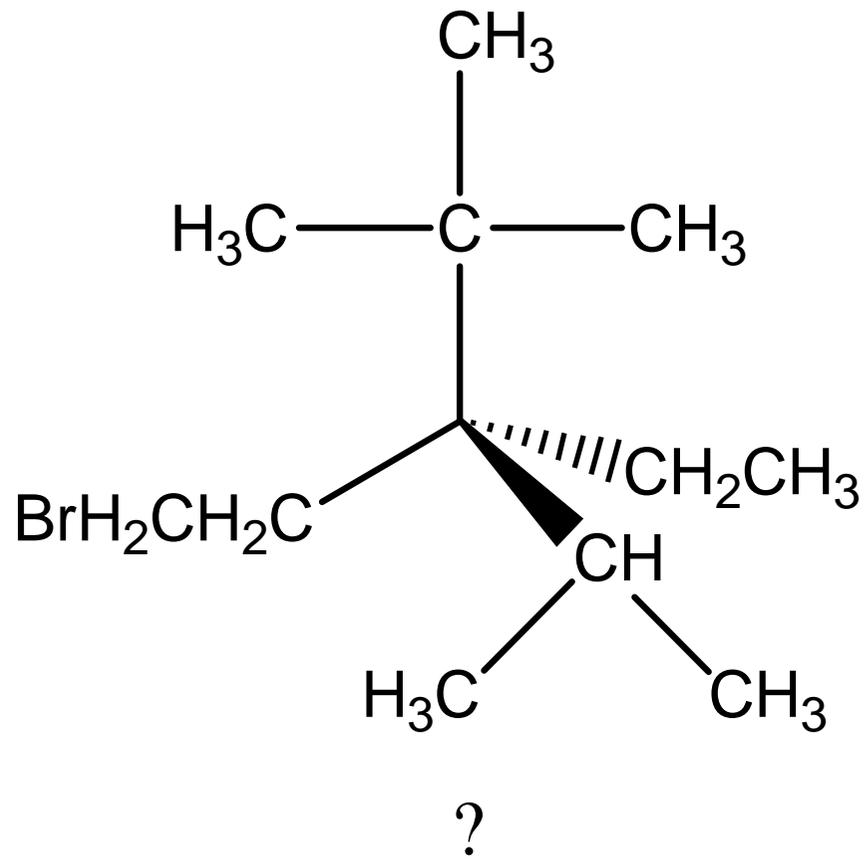
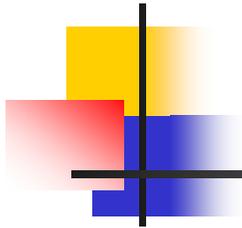
如:



R

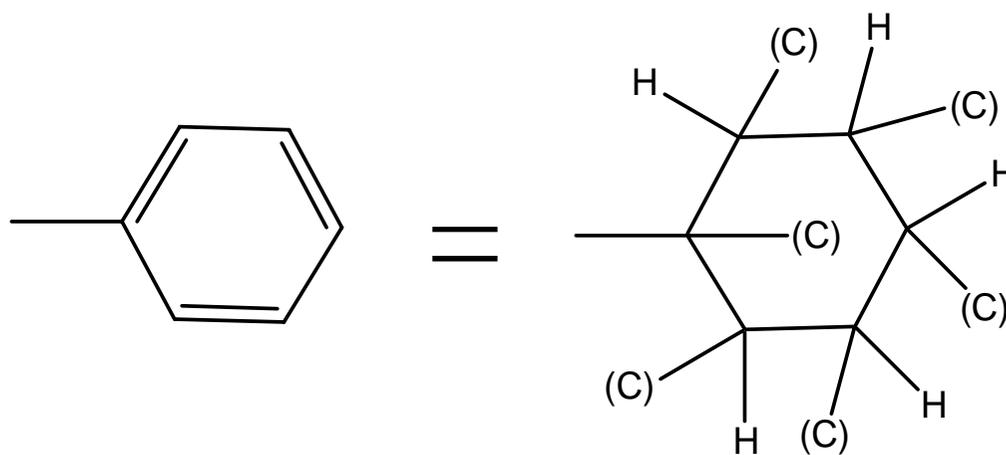
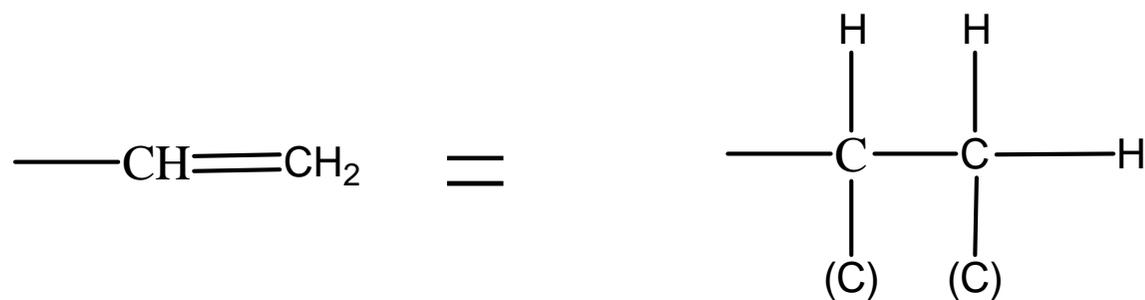
一个较大的原子序数比多个较小的原子序数优先

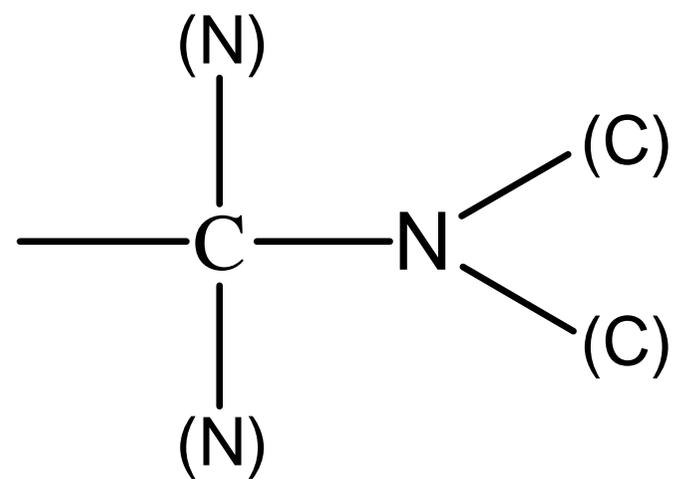
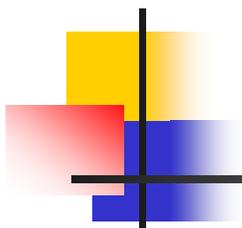


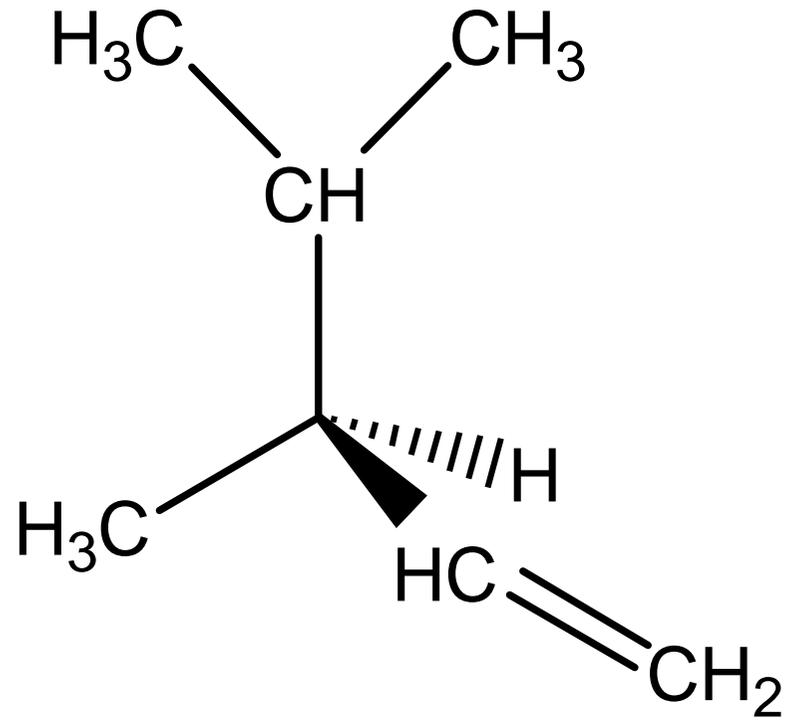
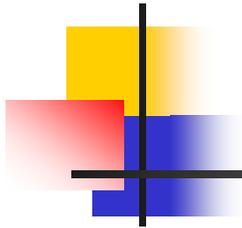


4, 若一个手性碳上连有双键或三键, 可将其拆开后再比较。

如:



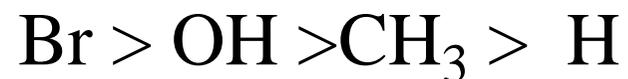
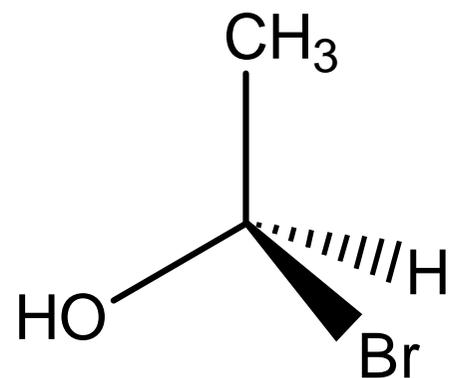


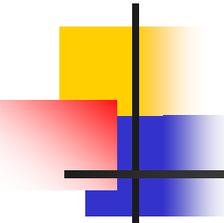


?

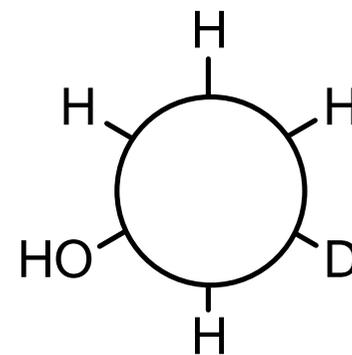
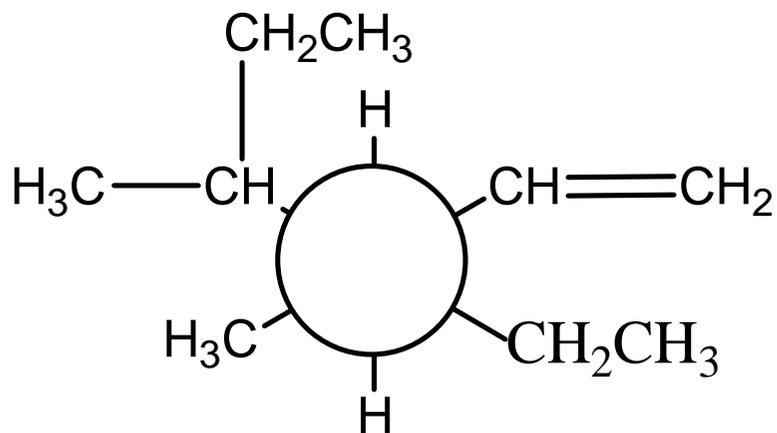
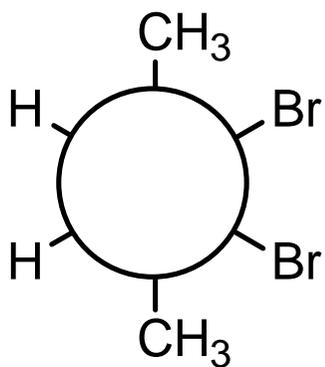
§ 3.4 构型在各种表达式中的标识

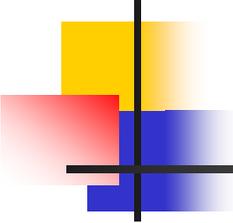
1, 构型在飞楔式中的标识:



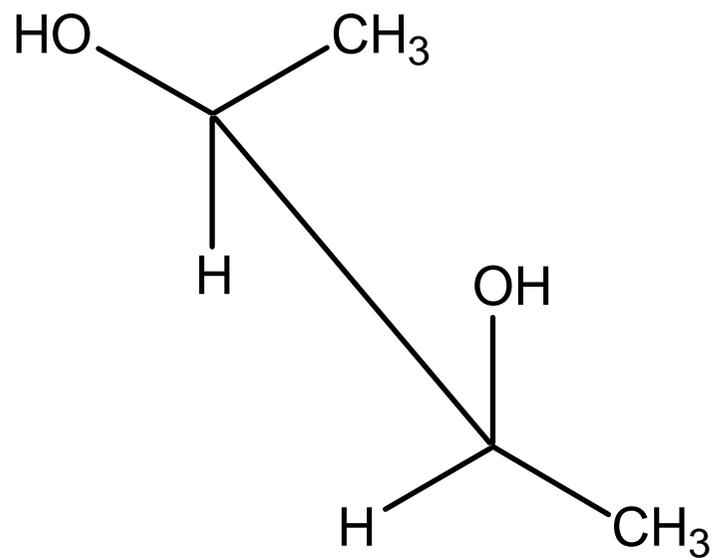
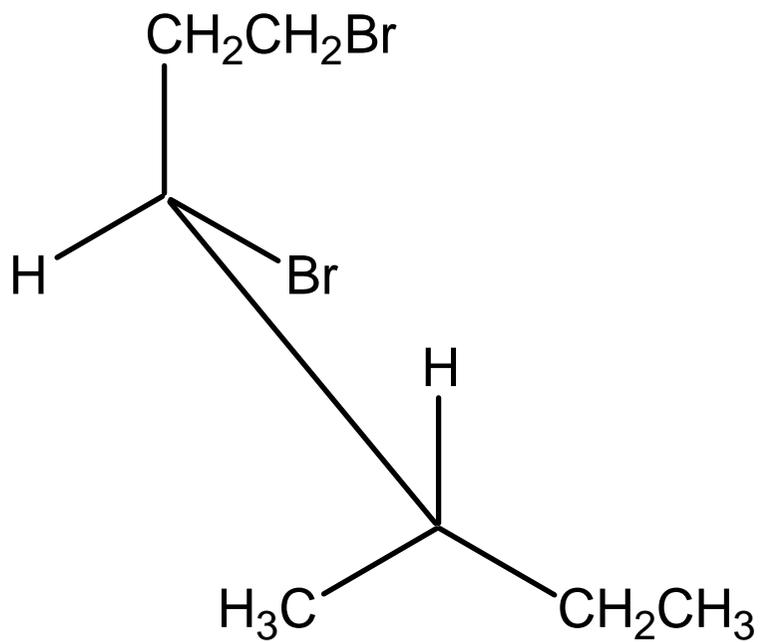


2, 构型在Newman投影式中的标识:

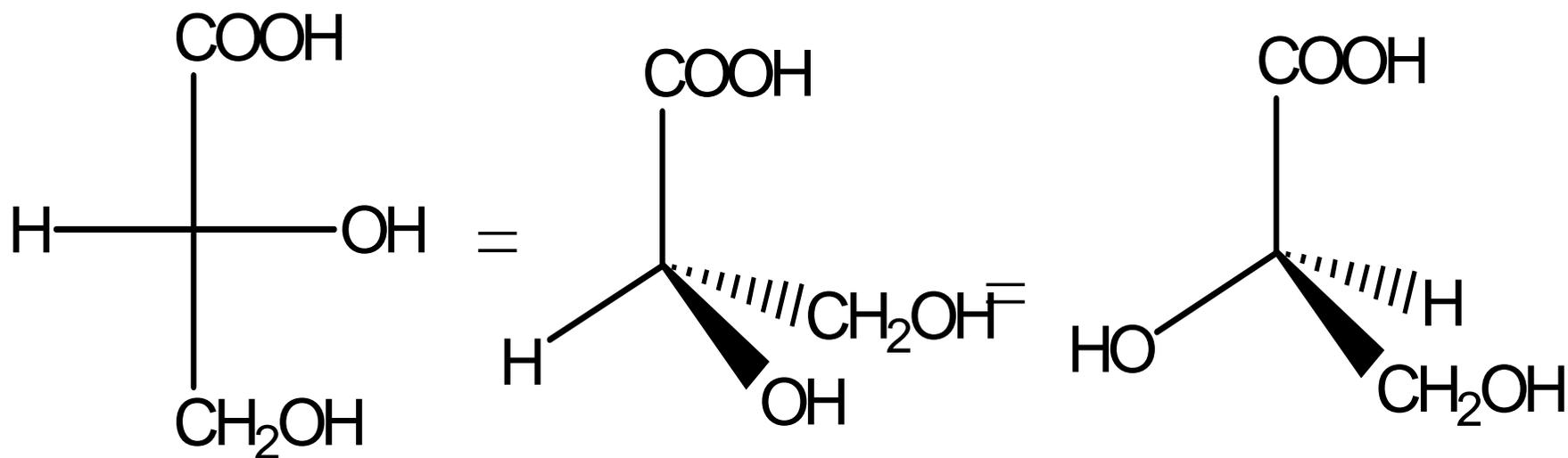




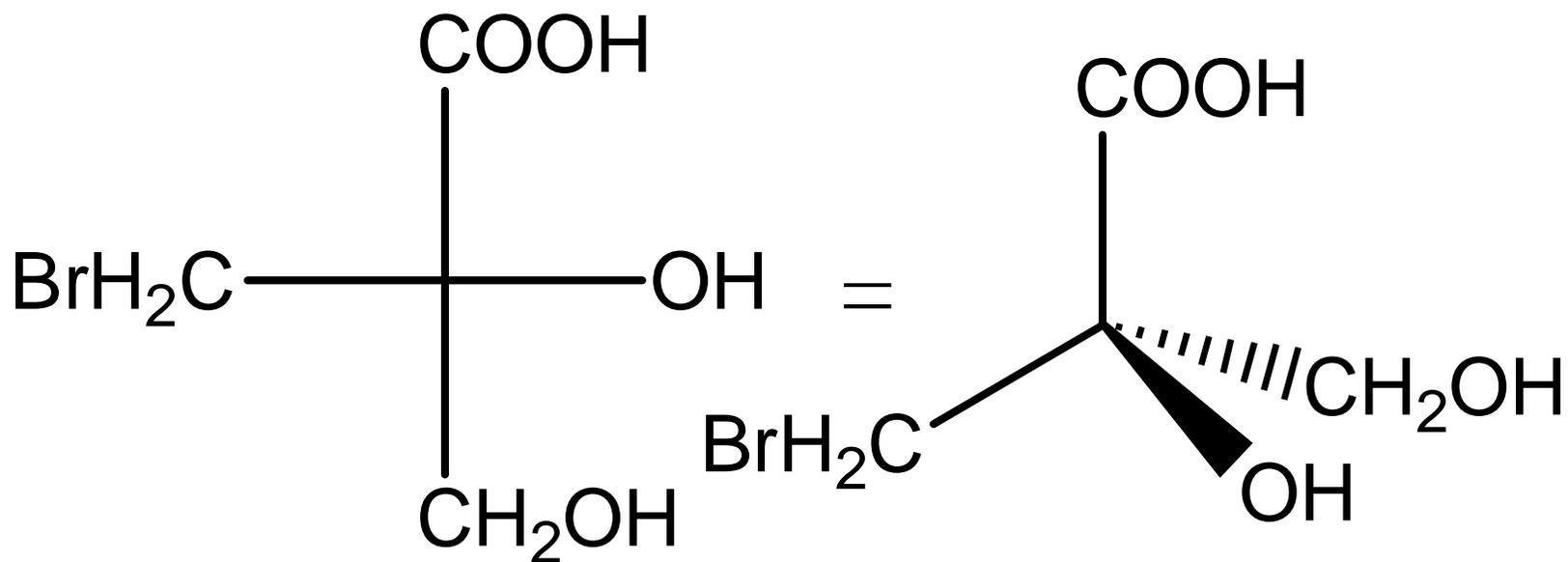
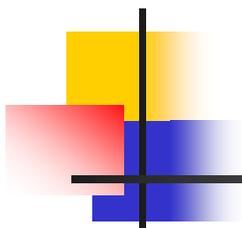
3, 构型在锯架式中的标识:

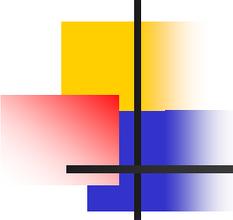


4, 构型在Fischer投影式中的标识:



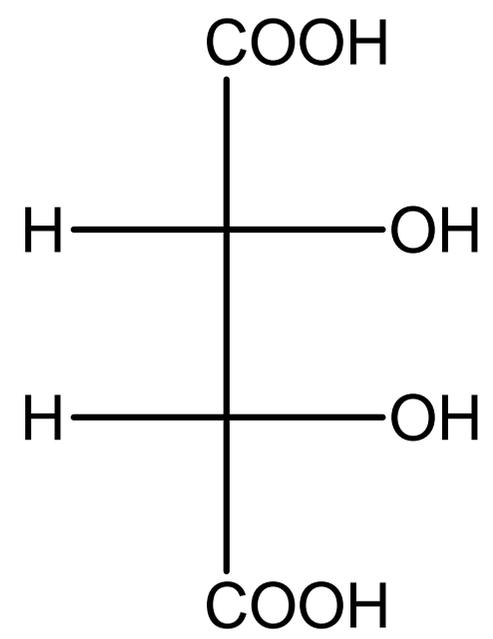
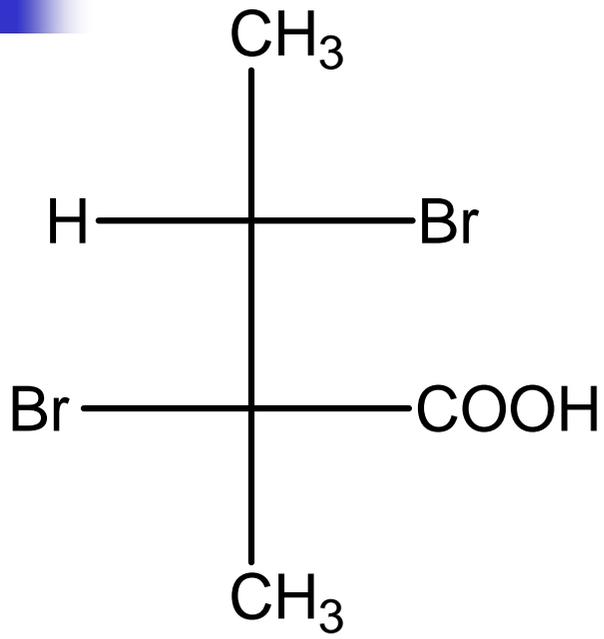
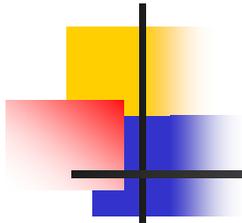
D-甘油酸

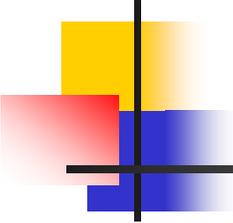




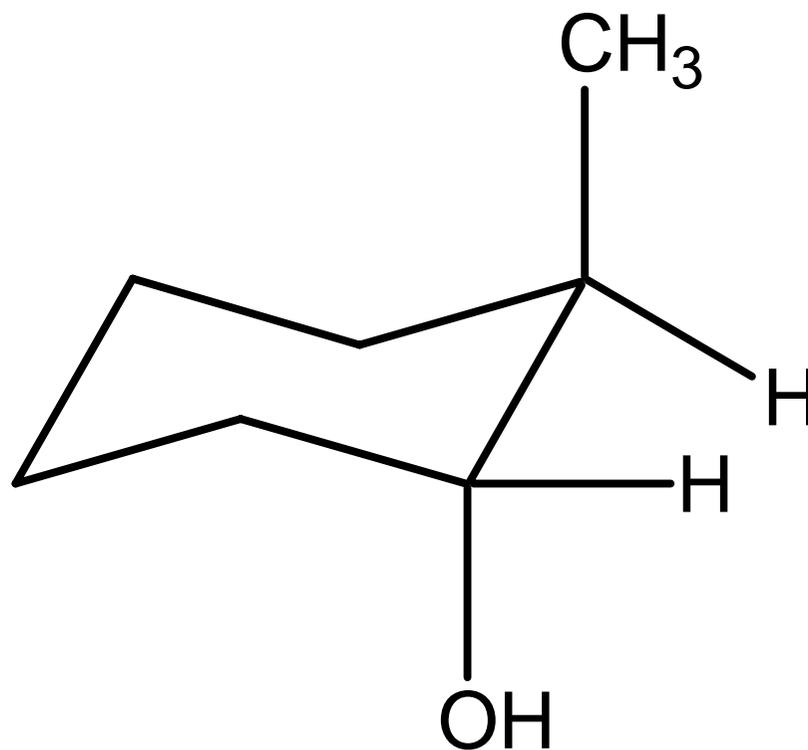
总结：

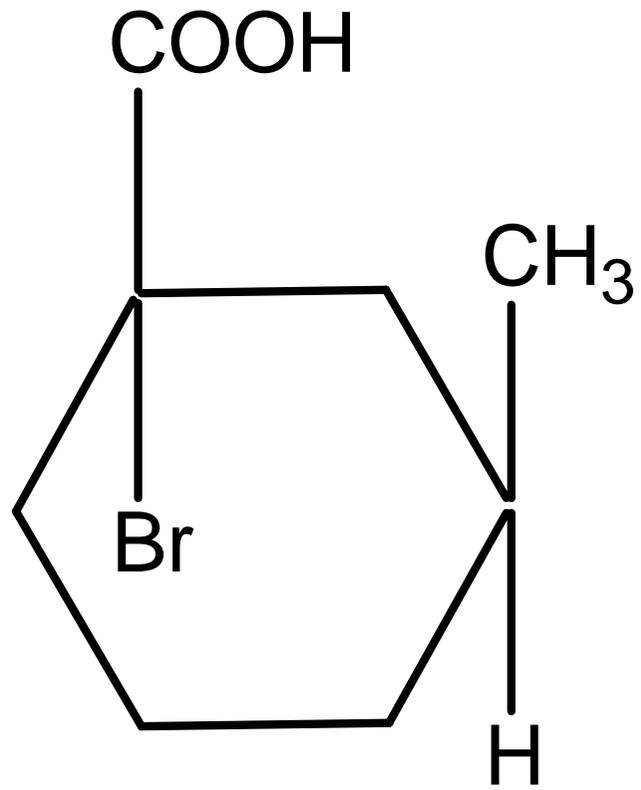
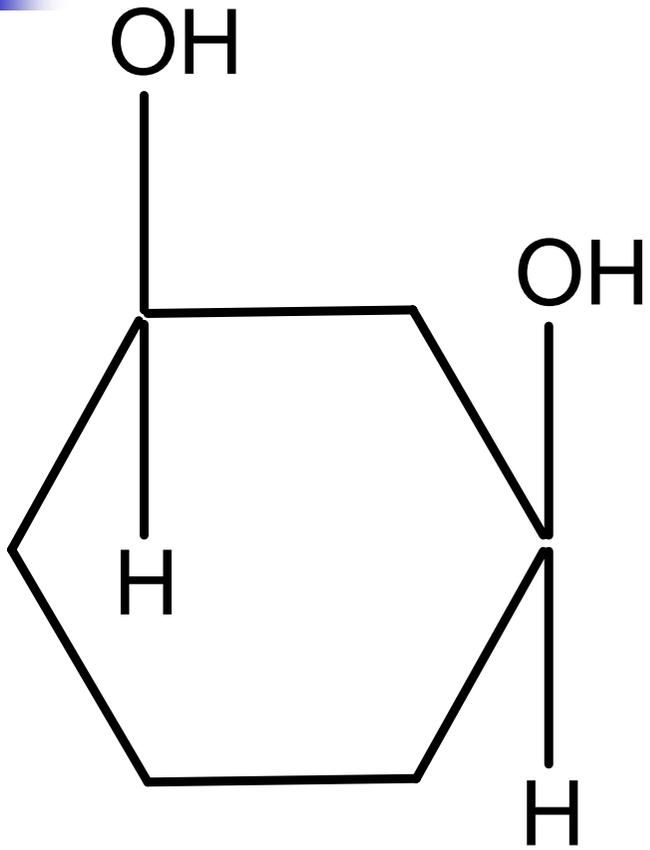
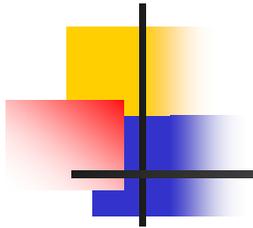
- A. 在Fischer投影式中，当最小基团在横键时，在Fischer投影式中的标定方向与实际标定方向相反，即顺时针则定义该手性碳的构型为 S 构型；反之，若为逆时针则定义该手性碳的构型为 R 构型。
- B. 在Fischer投影式中，当最小基团在竖键时，在Fischer投影式中的标定方向与实际标定方向相同。

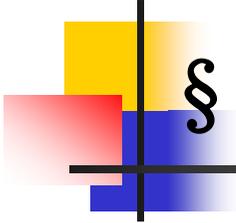




5, 构型在其它表达式中的标识:

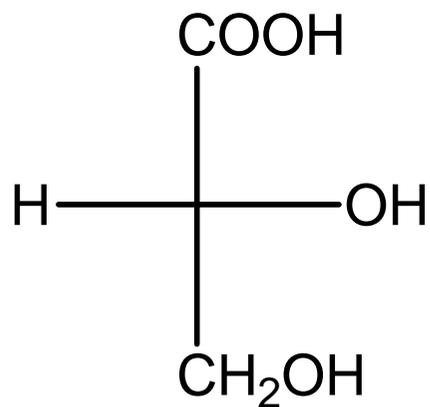




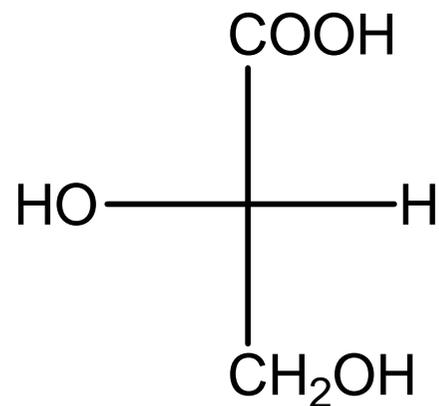


§ 3.5 手性化合物的旋光性及旋光度的测量

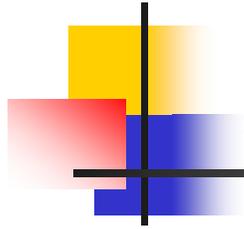
■ 1, 手性化合物的旋光性



(-)-甘油酸



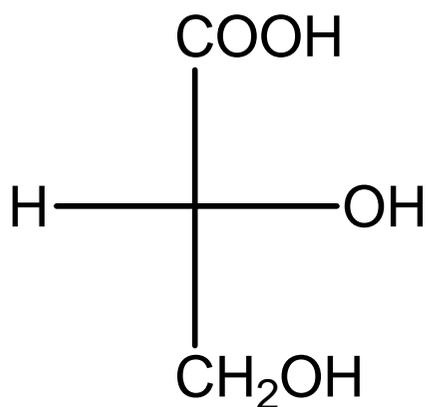
(+)-甘油酸



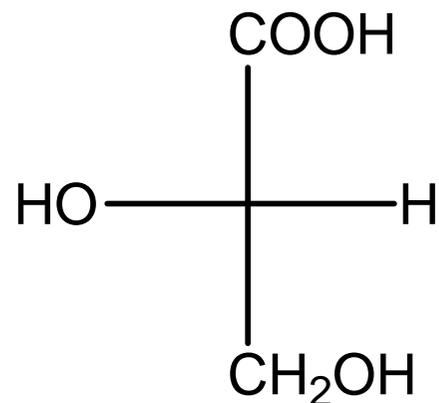
2, 旋光度的测量

§ 3.6 外消旋体及内消旋体

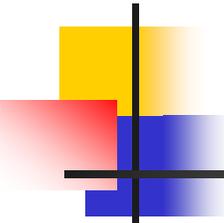
- 1, 外消旋体：等量的对映异构体的混合物



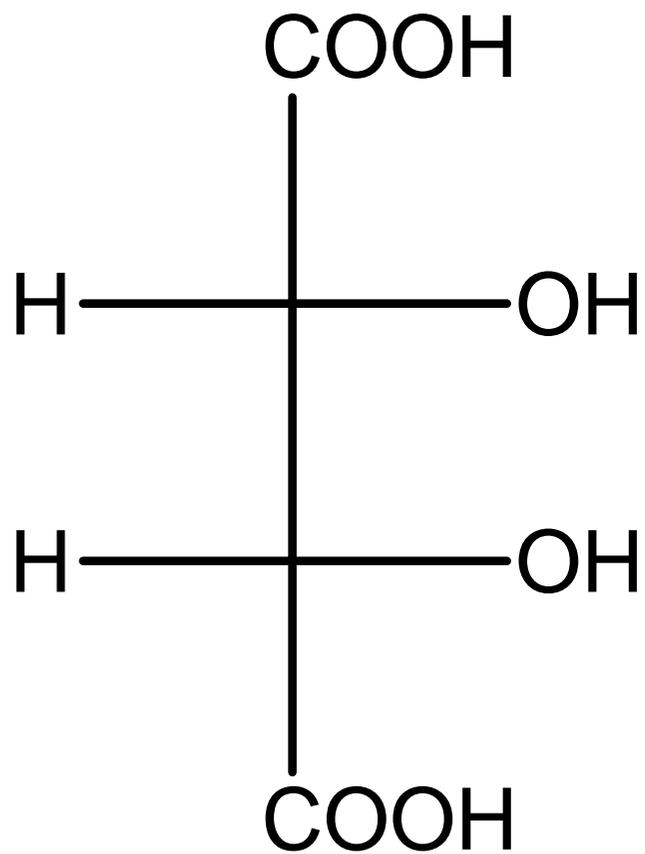
(-)-甘油酸

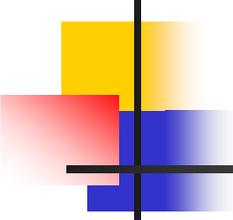


(+)-甘油酸



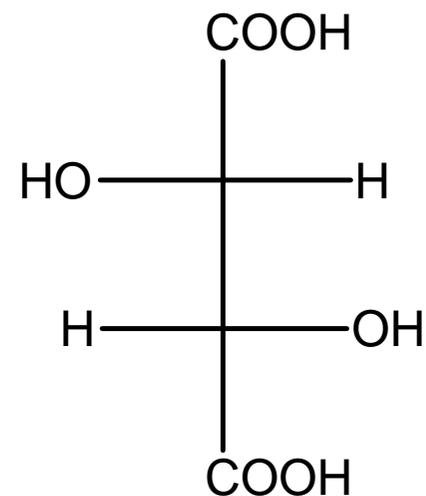
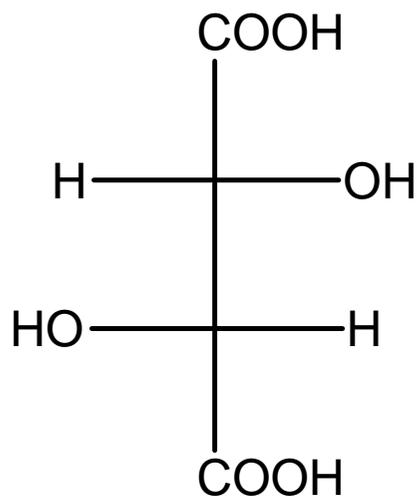
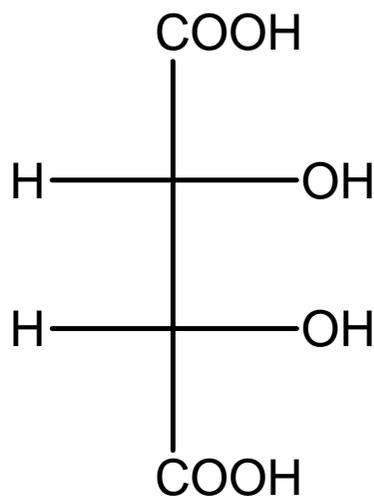
2, 内消旋体:

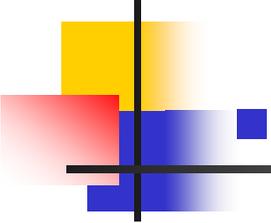




§ 3.7 含有两个手性碳的化合物

■ 1, 含有两个相同手性碳的化合物





2, 含有两个不相同手性碳的化合物

